

Available online at www.sciencedirect.com

Tetrahedron

Tetrahedron 63 (2007) 7753–7808

Tetrahedron report number 806

Recent advances in the chemistry of α , β -unsaturated trifluoromethylketones

Sergey V. Druzhinin, Elizabeth S. Balenkova and Valentine G. Nenajdenko*

Moscow State University, Department of Chemistry, Leninskie Gory, Moscow 119992, Russia

Received 30 March 2007 Available online 19 April 2007

Contents

Abbreviations: Ac, acetyl; (R)-Binol, (R)-(+)-1,1'-bi-2-naphthol; COD, 1,5-cyclooctadiene; DABCO, 1,4-diazabicyclo[2.2.2]octane; Dba, dibenzylideneacetone; DBU, 1,8-diazabicyclo[5.4.0]undec-7-ene; DDQ, 2,3-dichloro-5,6-dicyano-1,4-benzoquinone; Dess–Martin reagent, 1,1,1-tris(acetyloxy)-1,1-dihydro-1,2-benziodoxole-3-(1H)-one; DIP-Cl, (-)-B-chlorodiisopinocampheylborane; DME, dimethoxyethane; DMF, N,N-dimethylformamide; DMF-DMA, N,N-dimethylformamide dimethylacetal; DMSO, dimethylsulfoxide; fod, 6,6,7,7,8,8,8-heptafluoro-2,2-dimethyl-3,5-octanedione; LDA, lithium diisopropylamide; LTMP, lithium 2,2,6,6-tetramethylpiperidide; MCPBA, m-chloroperbenzoic acid; MsCl, methanesulfonylchloride; PMB, 4-methoxybenzyl; PPA, polyphosphoric acid; PPTS, pyridinium p-toluenesulfonate; Py, pyridine; TBAF, tetrabutylammonium fluoride; TBS, tert-butyldimethylsilyl; TFA, trifluoroacetic acid; TFAA, trifluoroacetic acid anhydride; Th, thienyl; THP, tetrahydropyranyl; TMSCF₃, trimethyl(trifluromethyl)silane; TMSCl, chlorotrimethylsilane; TMSCN, trimethylsilyl cyanide; p-Tol, p-tolyl (4-methylphenyl); Tos, 4-methylphenylsulfonyl; p-TsOH, p-toluenesulfonic acid. * Corresponding author. Fax: +7 495 932 88 46; e-mail: nen@acylium.chem.msu.ru

1. Introduction

The trifluoromethyl group, due to its unique stereoelectronic properties, is one of the most important substituents in organic chemistry. Because of the close Van-der-Waals radii $(CF_3=1.35 \text{ Å}, CH_3=1.29 \text{ Å})$, a compound containing a trifluoromethyl group is comparable with its methyl analog and this plays a very important role in drug–receptor interactions. The high electronegativity of the CF_3 group (3.5 on the Pauling scale) results in a quite different electron-density distribution and significantly changes the reactivity of the molecule. The influence of the trifluoromethyl group on the physiological activity is usually concerned with the increasing lipophilicity, which this substituent is bringing to the active molecules. The improvement of their transport characteristics in vivo and also the durability of the C–F bond compared with the C–H bond (116 and 100 kcal/mol, respectively), which allows the avoidance of undesirable metabolic transformations should be taken into consideration. In this respect, the introduction of trifluoromethyl groups into bioactive molecules, especially in the positions responsible for their physiological profile, becomes a very important direction in pharmaceutical studies that stimulates work directed to elaboration of synthetic methodology for various compounds containing trifluoromethyl groups.

The existing methods for direct fluorination and trifluoromethylation of organic compounds do not always allow the introduction of a CF_3 group in the required position of the molecule. As a result, a more flexible synthetic approach based on the application of simple and available fluorinecontaining compounds is a good supplement for direct fluorination methods and is nowadays gaining importance.

 α , β -Unsaturated trifluoromethylketones are readily available compounds. Their application in synthesis broadens the arsenal of existing building blocks and demonstrates the perspective and potential of this approach for the introduction of trifluoromethyl groups into compounds from different alicyclic, carbo- and heterocyclic classes.

The literature up to 1999 have been highlighted previously by members of our scientific group in two reviews^{[1,2](#page-49-0)} describing the synthesis of α , β -unsaturated trifluoromethylketones

and their application in organic synthesis. In the present review, the publications in this area over last seven years are discussed.

2. Methods for synthesis of α , β -unsaturated trifluoromethylketones

Analysis of the literature shows that several basic methods for the synthesis of α , β -unsaturated trifluoromethylketones were at the center of various investigations and these can be classified starting from the structure of the target compounds 1.

$$
F_3C_2\begin{array}{c|c} & R^2\\ \n 1 & 3 & R^3\\ \n 1 & R^1 & 1\end{array}
$$

— Creation of $C^2 - C^3$ bonds (e.g., trifluoroacylation of alkenes).

— Creation of $C^3 = C^4$ bonds (e.g., condensation of carbonyl compounds with 1,1,1-trifluoroacetone derivatives).

— Creation of $C^4 - R^3$ or $C^4 - R^2$ bonds (e.g., nucleophilic substitution at the β -carbon atom in α , β -unsaturated trifluoromethylketones containing a good leaving group in the β -position, reactions of amines with 1,3-dicarbonyl compounds, etc.).

— Creation of C^1 – C^2 bonds (e.g., trifluoromethyl-organometallic reagent addition to acrylic acid esters).

— Creation of $C^3 - R^1$ bonds (e.g., halogenation of α, β unsaturated trifluoromethylketones at the α -position).

The approaches for the synthesis of α , β -unsaturated trifluoromethylketones are described in the sequence given above.

2.1. Creation of C^2-C^3 bonds

Activated alkenes and compounds from which they are generated in situ can be trifluoroacetylated using trifluoroacetic acid anhydride (TFAA). This method is widely used, due to its simplicity and adaptability for a wide range of substrates. Nowadays, in spite of the fact that this method is well known, several groups still continue to investigate it.

Trifluoroacetylation with TFAA in the presence of pyridine was applied for the synthesis of 1,1,1-trifluoro-4-(hetaryl) but-3-en-2-ones 2. [3](#page-49-0) The starting materials were dimethyl acetals 3 of the corresponding ketones. The use of two equivalents of trifluoroacylating agent is needed. The first is consumed for generation of the enol ether, which is then trifluoroacetylated with the second equivalent of TFAA. The products were obtained as a mixture of E- and Z-isomers with *E*-isomer predominating.

(CF3CO)2O, Py, CHCl3 **²** (84-87%) **³** 0 °C, 4 h X = S, O ^X OMe Me OMe X OMe CF3 O E:Z=4:1 -10:1

The same methodology—the trifluoroacylation of dimethyl acetals 4 of ketones—was used for the synthesis of 1,1,1 trifluoro-4-methoxy-4-alkyl-3-buten-2-ones 5. [4](#page-49-0) It should be mentioned that, in the case of substrates containing bulky alkyl substituents, a higher temperature of ~ 60 °C is needed, while, using non-bulky alkyl substituents, room temperature is sufficient. The geometrical isomerism (E/Z) for compounds 5 has not been determined. The compounds 5 are highly volatile at room temperature and, because of this property, the elemental analysis of 5 could not be performed.

Ketone O-vinyl oximes, which have become accessible in recent times may also be regarded as O-substituted vinyl ethers. The goal of this work^{[5](#page-49-0)} was to obtain a deeper insight into specific features of the reaction of aliphatic and aromatic ketone O-vinyl oximes with TFAA. There are several examples of the trifluoroacylation of O-vinyl oximes 6 of various ketones forming the corresponding trifluoromethylketones 7. The reaction is stereoselective and only one more stable trans-isomer is formed. The trifluoroacylation of benzophenone O -vinyl oxime 8 was also investigated.^{[6](#page-49-0)} The reaction proceeds at room temperature for several hours and the target product 9 was isolated in low yield.

Besides TFAA, trifluoroacetyl chloride and bromide 10 (Hal=Cl or Br) have been used^{[7,8](#page-49-0)} for trifluoroacylation of 1,1-dichloroethene 11. The low activity of the double bond in 1,1-dichloroethene 11 requires the use of a catalyst. Aluminum halides were used as the catalysts in this reaction. Noteworthy, using aluminum bromide with 10 (Hal $=$ Cl or Br) the only product, 2,2-dibromovinyltrifluoromethyl-ketone 13, is formed,^{[8](#page-49-0)} whereas AlCl₃ with 10 (Hal=Cl or Br) gives the corresponding dichloroketone 12.

In order to prepare a number of β -alkoxyenones with various fluoro-containing substituents of different length and branching, as well as a different number of fluorine atoms, the reaction of polyfluorocarboxylic acid chlorides with ethyl vinyl ether instead of acid anhydrides was investigated.[9](#page-49-0) Various acyl chlorides 14 were applied for the acylation of 15 in the presence of pyridine in methylene chloride. The corresponding β -ethoxyketones 16 containing various fluorinated substituents were obtained in good yields.

$$
\begin{array}{c|c}\n\hline\n\text{OEt} & + & \text{O} & \text{CH}_2\text{Cl}_2, \text{Py, -10 °C, 2 h} \\
\text{15} & 14 & \text{then 20 h ft} \\
\text{16} & & 16 (40-83\%) \\
\text{R}_{f} = \text{CF}_{3}, \text{CHF}_{2}, \text{C}_{2}\text{F}_{5}, \text{C}_{3}\text{F}_{7}, \text{CH}(\text{CF}_{3})_{2}\n\end{array}
$$

In addition, the acylation of 2-methoxypropane 17 with anhydrides or perfluoroacyl chlorides and fluorides in the presence of triethylamine or pyridine has been described.[10](#page-49-0) Regardless of the carbon chain in the acylating agent, the target ketones 18 were obtained in high yields.

Me	O	Me		
OMe	R_f	X	R_f	OMe
17	18 (45-76%)			
R_f = CF_3, C_2F_5, C_4F_9, C_8F_{17}; X = CI, F, OCOR_f				

The behavior of several N-protected prolines 19 in the Dakin–West reaction was studied.^{[11](#page-49-0)} In the case of the application of TFAA as the acylating agent, instead of the target aminoketones the main products were the corresponding cyclic enaminoketones 20, which were obtained in moderate yields.

Furthermore, an example of CF_3 -enaminoketone formation in the acylation of a tertiary amine has been described. Enaminoketone 22 was obtained in low yield in the reaction of the substituted iso-tryptamine 21 with TFAA in the pres-ence of pyridine.^{[12](#page-49-0)} The reaction is not preparative, because of the low rate and the difficulty of isolation of the target product.

H

The application of TFAA and salen-manganese complex 23 in combination was described for the acylation of N-tosyl-2 pyrroline 24 in the presence of 2,6-di-tert-butyl-4-methylpyridine.[13](#page-49-0) The corresponding cyclic enaminoketone 25 was obtained in moderate yield.

On treatment with acetic anhydride, 5-hydroxypyrrolidin-2 one derivative 26 is converted into enamine 27, while, under treatment with TFAA, the formation of the heterocyclic enaminoketone 28 is observed, probably due to the more electrophilic character of TFAA.^{[14,15](#page-49-0)}

In addition to TFAA, trifluoroacetylimidazole 29 was used for the trifluoroacylation of cyclic enamines.[16](#page-49-0) Enamine 30 obtained from cyclohexanone and pyrrolidine was trifluoroacetylated at room temperature to form the corresponding enaminoketone 31 in moderate yield.

A similar approach was used for the synthesis of potato glycoalkaloid derivatives.¹⁷ Enamine 33 generated in situ from N-oxide 32 was trifluoroacetylated with TFAA, forming enaminoketone 34 in moderate yield.

The complex 35 prepared from 4-dimethylaminopyridine and trifluoroacetic acid anhydride was used for the trifluoroacetylation of various enamines 36 .^{[18](#page-49-0)} The reaction was carried out in benzene or methylene chloride as the solvent at room temperature. The yields of the products 37 are rather high, but the procedure seems to have no advantages over other methods of trifluoroacylation in the presence of other amines such as pyridine.

$$
\begin{array}{r|l|l} & R_1 & R_3 & \text{Me-S}_1\text{/Me} \\ \hline & R_1 & R_2 & \text{N} \\ & R_2 & \text{N} & 10\text{-}20\text{°C, }15\text{ h} & F_3\text{C} \\ \end{array} \hspace{1.2cm} \begin{array}{r|l|l} & R_1 & R_2 & \text{R}_2 \\ & R_1 & 37 & \text{R}_1 \\ & 35 & \text{C} & R_1 & 37 \\ & 36 & \text{C} & R_3\text{R}_4 = -(CH_2)_4-, -(CH_2)_2\text{O} \\ \text{R}_1\text{R}_2 = -(CH_2)_4-, (80\%)\text{; R}_1 = \text{Me}, R_2 = \text{Et}, (60\%)\text{;}\\ & R_1 = H, R_2 = t\text{-Bu}, (69\%)\text{; R}_1 = H, R_2 = \text{Ph}, (70\%)\text{;}\\ & R_1 = H, R_2 = 4\text{-}MeC_6H_4, (72\%)\text{; R}_1 = \text{Me}, R_2 = \text{Ph}, (75\%) \\ \end{array}
$$

The preparation of an unusual disubstituted $CF₃$ derivative of fulvene 38 has been described.[19](#page-49-0) This compound was obtained by trifluoroacylation of cyclopentadiene 39 with TFAA at room temperature using dimethylformamide as the solvent.

For the preparation of various ketones, the method based on cross-coupling reactions of thioesters 40 with the corresponding boronic acids 41 was used. The application of this method has been demonstrated for the synthesis of β -aryl-CF₃-enones, an example being ketone 42.^{[20](#page-49-0)}

$$
R_2
$$
\n
$$
R_2
$$
\n
$$
R_3
$$
\n
$$
R_4
$$
\n
$$
R_5
$$
\n
$$
R_6
$$
\n
$$
R_7
$$
\n
$$
R_8
$$
\n
$$
R_9
$$
\n
$$
R_1
$$
\n
$$
R_1
$$
\n
$$
R_2
$$
\n
$$
R_2
$$
\n
$$
R_1
$$
\n
$$
R_2
$$
\n
$$
R_2
$$
\n
$$
R_3
$$
\n
$$
R_4
$$
\n
$$
R_5
$$
\n
$$
R_6
$$
\n
$$
R_7
$$
\n
$$
R_8
$$
\n
$$
R_9
$$
\n
$$
R_1
$$
\n
$$
R_2
$$
\n
$$
R_3
$$
\n
$$
R_4
$$
\n
$$
R_5
$$
\n
$$
R_6
$$

 $R_1, R_2 = 3-NO_2C_6H_4, CF_3$; $R_1, R_2 = 3-NO_2C_6H_4, CF_3$; $R_1, R_2 = Ph$, $2-MeOC_6H_4$; $R_1, R_2 = Ph$, 3-MeOC₆H₄; $R_1, R_2 = Ph$, 2-Naph; $R_1, R_2 = Ph$, 4-HOC₆H₄; R_1 , R_2 = ClCH₂, 2-Naph; R_1 , R_2 = 4-NO₂C₆H₄, 3,4-methylendioxyphenyl

The method using cross-coupling of N-(4-methylphenyl)trifluoroacetimidoyl iodide 43 with diethyl allylphosphonate

44 in the presence of a palladium catalyst forming N-(4 methylphenyl)imino-5,5,5-trifluoropent-2-enylphosphonate 45 was elaborated. Compound 45 readily undergoes migration of the double bonds, isomerizing into compound 46. The structure of 46 was confirmed by ${}^{1}H$ NMR data. Additionally, the compound 46 can be used in the Horner–Emmons reaction with various aromatic aldehydes, generating the corresponding E,E-dienones 47 after acidic hydrolysis of the imine fragment in the imines $48.^{21}$ $48.^{21}$ $48.^{21}$

An analogous method was suggested for the preparation of enaminoketones 54 by acylation of imines 55 with ethyl trifluoroacetate. Imines 55 were prepared from substituted o -hydroxyacetophenones and various aliphatic amines.^{[23](#page-50-0)} The ketones obtained exist in the Z-form containing an intramolecular hydrogen bond.

Some CF_3 -enones were used for the preparation of effective catalysts for ethylene polymerization. The first neutral Ni(II)

OH N

 $R₂$

The possibility for introduction of a trifluoroacetyl group to the carbonyl group in various imines of methylketones 49 by reaction of the corresponding lithium azaenolates with Nsubstituted trifluoroacetimidoyl chlorides 50 or trifluoroacetic acid esters was investigated.[22](#page-50-0) In the first case, the products are diimines of diketones 51 . Using ¹H and ¹⁹F spectroscopy, it was established that the compounds exist in the form of tautomer 51b containing an enolized iminofragment neighboring trifluoromethyl group, so that the formed double bond exists in the Z-configuration. Only hydrolysis of the imino-fragment neighboring trifluoromethyl group is observed under acidic treatment for compounds 51. This leads to the formation of the β -enaminoketones 52.

LDA (2.0 eq.) THF, -78 °C NH4Cl (aq.) **51** (40-93%) N Cl R1 N Me R2 R3 F3C R3 ^N ^N R1 R2 Li F3C R3 ^N ^N R1 R2 **50 49**

$$
F_3C
$$

\n F_3C
\n F_1
\n R_1
\n R_2
\n F_3C
\n F_4
\n R_1
\n R_2
\n R_3
\n F_4
\n R_3
\n $THF, 60^\circ C$
\n F_3C
\n F_3C
\n R_3
\n51b
\n R_2
\n52 (58-75%)
\n R_3
\n R_1
\n R_2
\n53
\n54
\n R_3
\n552 (58-75%)

 \overline{F}

The second alternative approach for the synthesis of 52 consists of trifluoroacylation of imine azaenolates generated from 49 with methyl and ethyl trifluoroacetates 53, which allows the enaminoketones to be obtained directly in one step.^{[22](#page-50-0)}

polyethylene at modest rates. Incorporation of bulky substituents or perfluoroalkyl groups in the backbone of the P,O chelates greatly accelerated the polymerization rates. Several CF_3 -enaminoketones for subsequent preparation of such complexes with several fluorine-containing ligands were synthesized. 24 24 24 Ligand 56 was prepared by displacement of the Et₂N group in ((diethylamino)methylene)-1,1,1,5,5,5hexafluoroacetylacetone (DAMFA) by 2,6-diisopropylaniline catalyzed by $FeCl₃$ as Lewis acid by analogy with the known chemistry of DAMFA. Following chromatographic purification, compound 56 was isolated in good yields and high purity. Alternatively, similar ketones can be prepared

catalysts for ethylene polymerization were reported in the 1980s. These catalysts were primarily based on modifications of the SHOP (Shell higher olefin process) systems and incorporated anionic phosphino-enolate ligands. Such systems normally provided low-molecular-weight linear

 R_1 = Me, H; R_2 = CH₂Ph, *i*-Pr, $(CH_2)_2$ OH

 $N_{\sim H}$ -O

55 54 (51-82%)

 R_2

$$
R_3^{N-R_2}
$$

\n R_3
\n R_3

by acylation of silver salts with imidoyl chloride derived from sterically hindered anilines. The enamines 57 were readily deprotonated with an excess of sodium hydride in THF to yield the corresponding sodium salts. In situ reactions of these sodium salts with $(PPh_3)_2Ni(Ph)Cl$ at 25 °C generated the desired Ni(II) complexes 58. These complexes are active for ethylene polymerization in the presence of an activator $(Ni(COD)_2$ or $B(C_6F_5)_3$ to sequester PPh₃. Moderately branched polyethylenes generally in the range of 35–55 branches per 1000 carbons are produced, consistent with expectations based on the presence of *ortho*-disubstituted aryl groups on nitrogen.

An example of using trifluoroacetonitrile as acylating agent has also been described. The enolate generated from 1-acetylcyclohexanol 59 by treatment with a magnesium amide derivative was reacted with trifluoroacetonitrile to form the b-enaminoketone 60 by isomerization of intermediate com-pounds.^{[25](#page-50-0)} For a period of four months, this β -enaminoketone cyclizes into furan-3-one derivative 61 (yield is not given). After standing for 2.5 years, furan-3-one derivative 61

OH **66 64** (80%)

OH

rearranges into enaminoketone 62. All attempts to accelerate these reactions were unsuccessful.

Using an analogous strategy, enaminoketones containing imidazole substituents 63 and 64 were obtained. Ethyl trifluoroacetate was used for acylation of the starting compounds 65 and 66 with lithium diisopropylamide (LDA) as the base.^{[26,27](#page-50-0)}

2.2. Creation of $C^3 = C^4$ bonds

The traditional method for the creation of $C^3 = C^4$ bond is the condensation of carbonyl compounds with 1,1,1-trifluoroacetone or its derivatives under the conditions of Knoevenagel reaction.

The synthesis of ketone 67 was described using condensation of substituted chalkone 68 with trifluoroacetone in the presence of piperidine as a base.^{[28](#page-50-0)} The low yields in such reactions are caused by the self-condensation of the trifluoroacetone.

The reaction of ethyl trifluoroacetoacetate 69 with benzaldehyde 70 has been described.²⁹ The target product was formed by refluxing of the starting compounds in toluene in the presence of piperidine with azeotropic separation of water. The product 71 was obtained in moderate yield as a 1:1 mixture of E- and Z-isomers.

In a similar way, in the reaction of ethyl trifluoroacetoacetate 69 with triethyl orthoformate, derivative 72 is formed, which reacts with urea generating β -enamidoketone 73 containing

 CF_3

O

HO H_2N

an ethoxycarbonyl group in the α -position.³⁰ The yields for the obtained products are not given.

An original method for the synthesis of cyclic CF_3 -enones 74 has been elaborated. 31 The radical cyclization reaction of fluorinated 1,3-dicarbonyl compounds 75 with alkenes 76 was induced by manganese(III) acetate. Ketones 74 were obtained in various yields.

Several ketones 77 containing a β -dithiazolyl substituent have been obtained.^{[32](#page-50-0)} The starting compounds were fluorinated 1,3-dicarbonyl compounds 78 and 4,4-dichloro-1,2,3-thiazolium chloride (Appel salt) 79. A mixture of E,Z-isomers of 77 was formed.

Compounds 88 obtained by the Wittig reaction of phenylthio- or methylthiophenylphosphorane and ethyl trifluoroacetate can be easily lithiated by treating the free vinyl position with n-BuLi. The subsequent reaction with various aldehydes leads to the corresponding allylic alcohols 92. Under treatment with p-toluenesulfonic acid, compounds 92 are rearranged into the trifluoroketones 93.^{[36](#page-50-0)} All ketones 93 were obtained as their E-isomers. This method clears the

For a single example, the application of enamine 80, generated in situ from 1,1,1-trifluoroacetone 81 and pyrrolidine 82, for the synthesis of β -aryl-substituted CF₃-enone 83 by reaction with aromatic aldehyde 84 has been described.^{[33](#page-50-0)} The ketone 83 containing a 3-cyanophenyl substituent in the b-position was obtained in good yield. It should be noted that the method is suitable for the preparation of large amounts of ketone 83.

The synthesis of α , β -unsaturated trifluoromethylketones 85 using enaminophosphonates 86 in a Horner–Emmons reac-tion with various aldehydes has been described.^{[34,35](#page-50-0)} Compounds 85 can be prepared by condensation of diethyl alkylphosphonates 87 with trifluoroacetonitrile. Noteworthy is the fact that the trifluoromethyl group and the phosphorus atom in enamines 86 are located in the trans-position. In addition, all compounds 85 were obtained as their E -isomers.^{[34](#page-50-0)}

Eto-P

\nEto-P

\nEtO-P

\nEtO-P

\nIt is the following:

\n
$$
R_1 = H, Me
$$
\n
$$
R_2 = 4 - MeC_6H_4, 2-Th, 2-Fur, c-C_6H_{11}, 4-FC_6H_4, -8
$$
\n
$$
R_2 = 4 - MeC_6H_4, 2-Th, 2-Fur, c-C_6H_{11}, 4-FC_6H_4, -8
$$
\n
$$
R_2 = 4 - MeC_6H_4, 2-Th, 2-Fur, c-C_6H_{11}, 4-FC_6H_4, -8
$$
\n
$$
R_3 = 4 - MeC_6H_4, 2-Th, 2-Fur, c-C_6H_{11}, 4-FC_6H_4, -8
$$
\n
$$
R_4 = 4 - 4
$$
\n
$$
R_5 = 4 - 4
$$
\n
$$
R_6 = 4 - 4
$$
\n
$$
R_7 = 4 - 4
$$
\n
$$
R_8 = 4 - 4
$$
\n
$$
R_9 = 4 - 4
$$
\n
$$
R_1 = 4
$$
\n
$$
R_2 = 4 - 4
$$
\n
$$
R_3 = 4 - 4
$$
\n
$$
R_4 = 4
$$
\n
$$
R_5 = 4
$$
\n
$$
R_6 = 4
$$
\n
$$
R_7 = 4
$$
\n
$$
R_8 = 4
$$
\n
$$
R_9 = 4
$$
\n
$$
R_1 = 4
$$
\n
$$
R_2 = 4 - 4
$$
\n
$$
R_3 = 4
$$
\n
$$
R_4 = 4
$$
\n
$$
R_5 = 4
$$
\n
$$
R_6 = 4
$$
\n
$$
R_7 = 4
$$
\n
$$
R_8 = 4
$$
\n
$$
R_9 = 4
$$
\n
$$
R_1 = 4
$$
\n
$$
R_2 = 4 - 4
$$
\n
$$
R_3 = 4
$$
\n
$$
R_4 = 4
$$
\n

Several publications $36-38$ are devoted to application of substituted ethyl(1-trifluoromethyl)vinyl ethers 88–90 in the synthesis of α , β -unsaturated trifluoromethylketones. The ethers 88–90 were obtained using a Wittig reaction of ethyl trifluoroacetate and ylide 91.

way to preparing α -alkyl(aryl)thio- α , β -unsaturated trifluoromethylketones 93.

$$
\begin{array}{ccc}\n\text{RS} & \text{SR} & \text{SR} \\
\hline\n\text{SB} & \text{OEt} & \text{R}^1 \text{C} + 3 \\
\text{SB} & \text{OEt} & \text{R}^1 \text{C} + 3 \\
\text{SB} & \text{OEt} & \text{OH} & \text{OEt} & \text{R}^1 \text{O} \\
\text{R} = \text{Ph}, \text{Me}; \text{R}^1 = \text{Alk}, \text{Ar}, \text{Het} & \text{92 (58-90%)} & \text{93 (40-91%)} \\
\end{array}
$$

Compounds 89 were obtained by the condensation of bis- (phenylseleno)methane with ethyl trifluoroacetate with further elimination of the phenylselenyl group under treatment with methanesulfonyl chloride. The generation of the corresponding vinyllithium compounds is possible through two independent pathways—by lithiation of the unsubstituted vinyl position or by Se–Li exchange.[37](#page-50-0) In both cases, the organometallic derivatives obtained are used in the reactions with various aldehydes, forming the respective allylic alcohols 94 or 95. Subsequent acidic treatment leads to the α , β unsaturated trifluoromethylketones 96 or 97. In most cases, the reactions proceed stereoselectively. Compounds 96 are formed as the Z-isomers, excluding reaction with 3-phenylpropyn-2-al, in which case the E/Z-isomeric ratio in the α -phenylseleno- α , β -unsaturated trifluoromethylketone is 3:7. Compounds 97 are formed as the E -isomers.^{[37](#page-50-0)}

Vinyl selenide 90 was synthesized by condensation of phenylselenoacetonitrile with ethyl trifluoroacetate in the presence of methanesulfonyl chloride. Subsequent Mg–Se exchange using EtMgBr and treatment with 2,4,6-trimethylbenzaldehyde leads to the benzylic alcohol 98, which can be transformed into the α -cyano- α , β -unsaturated trifluoromethylketone 99 by reaction with p -toluenesulfonic acid.^{[38](#page-50-0)} It is to be noted that using n -BuLi as the reagent for the Li–Se exchange causes a decrease in the yield of 98 to 17%. This synthetic sequence opens up the pathway to α -cyano- α , β -unsaturated trifluoromethylketones 99.

The Claisen rearrangement can also be used for the synthesis of α , β -unsaturated trifluoromethylketones. A method for the preparation of CF_3 -dienones based on the application of this rearrangement has been elaborated.^{[39](#page-50-0)} 1-Phenylsulfanyl-2-bromo-3,3,3-trifluoropropene 100 served as the starting material. Its reaction with sodium hydride leads to in situ formation of the CF_3 -acetylene, which reacted with various allylic alcohols 101 to form the vinyl ethers 102. The vinyl ethers were rearranged by heating in carbon tetrachloride solution to generate the γ , δ -unsaturated ketones 103. The target dienone 105 was formed by oxidation of phenylsulfanyl group in 103 with m-chloroperbenzoic acid and elimination of the sulfinic acid from the sulfoxide 104. The stereochemistry of the obtained compound is explained by a more favorable chair-like conformation of transition state in the Claisen reaction and also by syn-elimination of sulfinic acid from the sulfoxide 104.

through the intermediate ester 108 and Michael addition of propargylic alcohol 109 with further elimination of fluoride anion from intermediate 110. Vinyl propargyl ether 106, in turn, by heating in toluene at 80 $^{\circ}$ C for 3-4 h undergoes Claisen rearrangement with further double-bond migration leading to 111 (Z/E ratio 2/1).^{[40](#page-50-0)}

The Claisen reaction was also studied for various fluoroalkyl-containing vinyl propargyl ethers. Compound 106 was prepared by subsequent dehalogenation of ester 107

vinyl propargyl ether has not been isolated—in the reaction conditions, it rearranges directly into the dienone (Z/E-ratio 1.2/1).

An attempt to carry out analogous transformations for 1,1 dimethylpropargyl alcohol 114 failed. In this case, preparation of dienone 115 using the reaction of acrylate 116 with alcohol 114 in tetrahydrofuran in the presence of sodium hydride was described. The ratio of Z/E isomers was 1/1 and only the E-isomer was isolated by chromatography.

2.3. Creation of C^4-R^3 or C^4-R^2 bonds

 α, β -Unsaturated trifluoromethylketones having a heteroatom (e.g., alkoxy or dialkylamino) in the b-position can take part in reactions with nucleophiles by an 'addition–elimination' mechanism with the further formation of new α , β -unsaturated trifluoromethylketones.

In addition, 1,3-dicarbonyl compounds containing a trifluoroacetyl fragment react easily with primary and secondary amines. The reaction proceeds as an addition on the carbonyl atom with further water elimination and preservation of the trifluoroacetyl group. The products of this reaction are β -enaminoketones containing a trifluoroacetyl group. A large number of reports are devoted to investigations of this reaction.

In one example, the reaction of cyclic 1,3-diketone 117 containing a CF_3 group with *o*-phenylenediamine 118 was studied.^{[41](#page-50-0)} The reaction proceeds rapidly under reflux in methanol solution in the presence of acetic acid. The corresponding enaminoketones 119 were prepared in low yields.

The analogous reaction of 1,3-diketones 75 with substituted o -phenylenediamine 120 is described.^{[42](#page-50-0)} Noteworthy is the fact that the nucleophilic attack of the more nucleophilic NH2 group is not directed onto the trifluoroacetyl group. The β -enaminoketones 121 were obtained in good yields.

Under reaction of *o*-phenylenediamine 118 with triketones 122 containing two fluorinated alkyl groups, the formation of the corresponding 1,5-benzodiazepines 123 is observed.⁴³ The reaction proceeds in good yields during 10–15 min of methanol reflux. Equilibrium of the two tautomeric forms for the obtained products 123 is observed.

The reactions of CF_3 -enone 124 with 2-aminothiophenol 126 and 2-aminophenol 125 were investigated. The reactions were carried out in toluene and the ketone 124 reacted easily to give 127 or 128 in high yield.^{[44](#page-50-0)}

The reaction of dihydropyran-4-one 129 with ammonia proceeds by an analogous route. Dihydropyran-4-one 129 was obtained by treatment of 1,3-dicarbonyl compound 130 with aqueous hydrogen chloride.^{[45](#page-50-0)} Reaction of 129 with ammonia was complete in 2 days at room temperature, but the yield of enaminoketone 131 was not very high.

Similarly, this reaction proceeds with the hydroxy-pyranone 132. CF_3 -enaminoketones 133 were obtained in good yields by treatment of 132 with aqueous ammonia.⁴⁶ Applying benzylamine instead of ammonia leads to acidic splitting of the 1,3-dicarbonyl compound with loss of the trifluoroacetyl group instead of enaminoketone 133 formation. The products of the reaction in this case were hydroxyketone 133a and N-trifluoroacetylbenzylamine 133b.

Ketones 134 containing an aziridine fragment in the β -position were prepared by the reaction of cis-1,2-diphenylaziridine 135 with CF₃-enone 136 containing a chlorine atom in the β -position. The reaction proceeds with the formation of a mixture of E,Z-isomers of ketone 134.^{[47](#page-50-0)}

A $BF_3 \cdot Et_2O$ complex was used to accelerate the reaction of 1,3-diketones 137 with aromatic amines and to increase the yields.[48](#page-50-0) Catalytic amounts of this complex essentially accelerate the reaction and increase the yields of the target enaminoketones 138.

In addition, zinc perchlorate hexahydrate was found to be an effective catalyst for this reaction. 49 This catalyst was very active for the transformation of 1,1,1-trifluoroacetylacetone into the enone 139 in the reaction with aniline.

$$
\begin{array}{c}\n0 & 0 \\
\hline\nM & CF_3\n\end{array} + \begin{array}{c}\nPhNH_2 \xrightarrow{Zn(CIO_4)_2 \cdot 6H_2O \ (5 \text{ mol\%})} \text{Ph} \cdot N \xrightarrow{H} O \\
\hline\nMgSO_4 \ (30 \text{ mol\%})\n\end{array} \begin{array}{c}\n\text{Ph} \cdot N \xrightarrow{H} O \\
\hline\nMgSO_4 \ (30 \text{ mol\%})\n\end{array}
$$

The preparation of the target enamine 140a by the reaction between diketone 141 and benzylamine using waterseparation conditions failed.^{[50](#page-50-0)} The yield of the expected enamine 140a in this reaction was only about 11%. Less nucleophilic trifluoroacetates of amines were used in an attempt to improve the yield of 140a. The reaction between diketone 141 and benzylamine trifluoroacetate proceeded at a good rate. The important feature of this reaction was the absence of unwanted by-products. To demonstrate the practicality of the developed reaction conditions using benzylamine trifluoroacetate as a reagent, the reaction was performed on a relatively large scale (20 g) with successful reproducibility in the yield of the enamine 140b.

The enones 142 can be used as selective protecting groups for the α -amino group of α -aminoacids 143 containing additional functional groups in the side chain (arginine, asparagine, etc.). The protection is provided by formation of the enaminoketone 144. An exception is lysine, in which case both amino groups react equally.^{[51](#page-50-0)} The cleavage of this protective group is performed by treatment with hydrogen chloride in methanol.

The reactions of 5-trifluoroacetyl-3,4-dihydro-2H-pyran 145 with various nucleophiles have been studied.^{[52](#page-50-0)} This ketone reacts readily with many nucleophiles such as amines

and Grignard reagents to give the ring-opened products 146. Hydrazine and hydroxylamine attack the carbonyl carbon of compound 145 to form hydrazone or oxime 147.

Additionally, enone 154 was used for the preparation of conjugated trifluoromethylenones 156 containing an acetylenic fragment.[56](#page-50-0) The reaction of enone 154 with various lithium

In recent years, environmentally benign synthetic methods have received considerable attention and new procedures for the synthesis of enaminones have been reported. The results of a gold(III)-catalyzed synthesis of β -enaminones 148 from 1,3-dicarbonyl compounds 149 have been reported.^{[53](#page-50-0)} This procedure is quite general for a wide range of amines such as aliphatic, cyclic and aromatic amines using the commercially available catalyst $NaAuCl₄·2H₂O$.

$$
R = Me, 2-Th
$$

\n
$$
R = H, R_2 = \frac{Me}{R_1}
$$

\n
$$
R_1 = H, R_2 = \frac{Me}{R_2}
$$

\n
$$
R_3C
$$

\n
$$
R_4 = H, R_2 = \frac{Me}{R_1}
$$

\n
$$
R_1 = H, R_2 = \frac{Me}{R_1}
$$

\n
$$
R_1 = H, R_2 = \frac{Me}{R_1}
$$

\n
$$
R_1 = H, R_2 = \frac{Me}{R_1}
$$

\n
$$
R_1 = H, R_2 = \frac{Me}{R_1}
$$

\n
$$
R_1 = H, R_2 = \frac{Me}{R_1}
$$

\n
$$
R_1 = H, R_2 = n-Bu
$$

Deoxofluor 150 appears in some cases to be a more effective fluorinating reagent than DAST 151 and is easier to handle than sulfur tetrafluoride. A new route to polyfluoro ethers from alkyl, aryl, and mixed glyoxal hydrates and other hydrates using Deoxofluor has been reported.⁵⁴ The reaction of Deoxofluor with hexafluoroacetylacetone dihydrate 152 led to the formation of the β -enaminoketone 153 in good yield, without the formation of either a cyclic or an acyclic ether. No vicinal difluoroamine formation was observed in the reaction, indicating that the condensation proceeds more rapidly, producing the β -enaminoketone as the main product.

This work^{[55](#page-50-0)} continues with an investigation of the application of CF_3 -enaminones for the synthesis of β -alkyl(aryl)enones. The reaction of various Grignard and organolithium reagents with enone 154 was investigated. The reaction proceeds stereoselectively, leading to the formation of only the trans-isomer of CF_3 -enone 155. It was shown that the use of organolithium reagents has no advantages over Grignard reagents. Application of latter in several cases is more justified. It was shown that the application of Grignard reagents modified with cerium(III) salts does not lead to any improvements, in comparison to the non-modified reagents.

$$
\begin{array}{ccccc}\n\text{Me}_{2}N & \longrightarrow & \text{CF}_{3} & \xrightarrow{R-M} & \text{R} & \longrightarrow & \text{CF}_{3} \\
\downarrow & & \downarrow & & \downarrow & & \downarrow & \\
 & 154 & & 2 h, \text{reflux} & & 155 (37-80\%) \\
 & M = MgBr; R = Alk, Ar & & \\
 & M = Li; R = Ph, Ph-C=C \cdot \frac{3}{2}\n\end{array}
$$

derivatives of terminal acetylenes leads to the formation of the target products in moderate yields. Diketones 157 were obtained in moderate yields using the dilithium derivatives of di-acetylenes.

The same reaction of organolithium derivatives with enaminoketones 158 was applied to the synthesis of cyclobutene ketones 159 containing various substituents in the β -position.^{[57](#page-50-0)} The formation of the corresponding hydroxyketones 160 as by-products was observed.

The problem of the formation of by-products—the corresponding keto-alcohols 160—was solved using the reaction of enaminoketones 161 with Grignard reagents. In this case, the reaction proceeds regioselectively to form the target CF_3 -enones 162 in high yields.^{[58](#page-50-0)}

A cross-coupling reaction for the synthesis of bicyclic cyclobutene ketones 163 containing various substituents in the β -position has been reported.^{[59](#page-50-0)} The reaction of arylzinc derivatives with the corresponding bromide 164 in the presence of $(Ph_3P)_4Pd$ catalyst was used.

The greater effectiveness of β -alkoxy-CF₃-enone 124 in the reaction of the lithium derivative of phenylacetylene 165 was demonstrated for preparing the trifluoromethyl enone 166 containing an acetylenic fragment.^{[60](#page-50-0)}

$$
Ph \xrightarrow{ \overline{H}} \underline{Li} \quad + \quad \overbrace{EtO}^{O} \xrightarrow{ \overline{C}F_3} \xrightarrow{ \overline{P}h \xrightarrow{ \overline{C}} \overline{P_3} \xrightarrow{ \overline{D}} \overbrace{C_{F_3}}^{O} \xrightarrow{ \overline{C}F_3} \overbrace{C_{F_3}}
$$

The reaction of ethoxyketone 124 was used for the preparation of β -uracil-substituted CF₃-enone **167** containing a sulf-imine group.^{[61](#page-50-0)} The reaction with imine 168 was performed in methylene chloride at room temperature.

A number of reports^{[62,63](#page-50-0)} are devoted to the application of 4-sulfonyl-1,1,1-trifluorobut-3-en-2,2-diols 169 for the synthesis of various β -amino- and β -thio- α , β -unsaturated trifluoromethylketones. Compounds 167 can be prepared in yields close to quantitative by oxidizing β -thiosubstituted enones 170 prepared by trifluoroacylation of the corresponding vinyl sulfides 171. [64](#page-50-0) The oxidation was carried out using 50% H₂O₂ aqueous solution in the presence of trifluoroacetic acid or utilizing 98% H_2O_2 in the presence of trifluoroacetic acid anhydride. Subsequent hydration of the 4-sulfonyl-1,1,1-trifluorobut-3-en-2-ones 172 formed in the reaction proceeded in quantitative yields. 63 In the case of compound 172 with R=Ph, the E/Z -isomeric ratio is 7/2, and with $R=Me$, only the E-isomer is formed.

The reaction of 169 with various amines 173 or their aqueous solutions leads to the formation of corresponding enaminoketones 174 in high yields.^{[61](#page-50-0)} When one of the substituents $(R¹$ or $R²$) is a hydrogen atom, the formation of the Z-isomer of 174 is observed (except when $R^2 = 3$ -Py). If the formation of an intramolecular hydrogen bond N-H \cdots O is impossible, the formation of the E -isomer is observed.⁶¹ The reaction is completed in the presence of only 1 equiv of amine, in spite of the evolution of sulfinic acid. This can be explained by the high reactivity of 169.^{[61](#page-50-0)}

In the same work, the influence on the Z/E-isomer ratio in 174 (R^1 =H, R^2 =Ar) of substituents in the benzene ring was investigated. The presence of electron-donating substituents stabilizes the Z-isomer more favorably, because of intramolecular hydrogen bond $N-H\cdots O$; the opposite situation is observed when an electronegative group is presented. Additionally, in nonpolar solvents (e.g., chloroform) the equilibrium is completely turned to the Z-isomer.^{[61](#page-50-0)} The configuration of the double $C = C$ bonds in compounds 174 was established using ¹H NMR spectroscopy.

The reaction of 169 with various thiols 175 demonstrates a new pathway to β -sulfanylenones 176.^{[63](#page-50-0)} The reaction proceeds in mild conditions in high yields. In most cases, the Eisomer is formed predominantly in this reaction. The yield of the product is not influenced by the nature of R substituent, except in the case of p -nitrothiophenol, where only the Z-iso-mer has been isolated.^{[63](#page-50-0)} The opposite correlation is observed for more nucleophilic p-methoxythiophenol 177: in addition to the normal product 178, the double-addition product 179 was isolated; using a twofold excess of *p*-methoxythiophe-nol 177 allows the product 179 to be obtained in high yield.^{[63](#page-50-0)}

In addition, the diol form of ketones 169 was used for the preparation of CF_3 -enones containing heterocyclic substituent in the β -position. This transformation was achieved using the reaction of 167 with triazole 180, imidazole 181, 3,5-dimethylpyrazole 182 and their benzoderivatives 183.^{[65](#page-50-0)} CF₃-enones 184–186 containing heterocyclic substituents were obtained in high yields.

The same workers^{[66](#page-50-0)} have reported the application of a new electrophilic reagent 187, obtained by the oxidation

of ketene dithioacetal 188. This reagent was applied for the synthesis of various α -phenylsulfonyl- α, β -unsaturated trifluoromethylketones. In one example, the reaction with 1,3-dimethoxybenzene leads to mixture of isomers 189 and 190 in a 7/1 ratio. The reaction of 187 with 2-methylthiophene proceeds 100% stereoselectively and only the E -isomer of ketone 191 is formed.^{[66](#page-50-0)}

The reaction of 187 with more nucleophilic substrates proceeds without elimination of sulfinic acid and leads to the formation of adducts 192 and 193 (a mixture of diastereomers is formed). Elimination of sulfinic acid can be easily provoked by reflux of a methylene chloride solution for compound 193, and using triethylamine as the base in the case of compound 192. [66](#page-50-0) Ketones 194 and 195 were obtained in high yields.

A method for the preparation of α , β -unsaturated ketones 196 containing an $-OTs$ leaving group in the α -position to the carbonyl group has been described.[67](#page-50-0) These compounds are formed under treatment of vinyl tosylate 197 with sulfuric acid in tetrahydrofuran. The allyllic alcohols 197 were prepared from lithium derivative 198 and carbonyl compounds. The intermediate lithium derivative 198 can be synthesized with isolation of fluoroalkene 199 and also by direct metallation of 200 with 2 equiv of *n*-BuLi. The target ketones 196 were obtained in good yields and high stereoselectivity.

Polyfluorinated aldehydes 201 were used as starting compounds for the synthesis of β -enaminoketones 202.^{[68](#page-50-0)} The target N-substituted β -enaminoketones 202 are formed in good yields by reflux of an acetonitrile solution of the polyfluorinated aldehydes with various amines in the presence of water.

The synthesis of β -selenoenones 203 using the reaction of methoxyenones 204 with methyl- and phenylselenol in the

EtO

 $CF₃$

 $\overline{1}$

 R_{F} = CHF $_2$, CF $_3$ \quad R $_2$ /R $_1$ = Ph/H, 4-MeC $_6$ H $_4$ /H, 4-MeOC $_6$ H $_4$ /H, 4-ClC $_6$ H $_4$ /H, 1-Naph/H, 2-Th/H, MeCH=CH/H, PhCH=CH/H, Et/Et *n*-Pr/H, *t*-Bu/H,

presence of 1 equiv of the BF_3 -diethyl ether complex has been described.^{[69](#page-50-0)}

The substitution of one alkoxy group in enone 205 was used for the synthesis of O,N-acetals-aminals of trifluoroacetylketene 206. The reaction was carried out in an aqueous medium. The method is characterized as being simple and effective and the yields of the products are rather high.^{[70](#page-50-0)}

$$
\begin{array}{ccc}\n & & \text{OEt} & \\
 & & \text{ORt} & \\
 & \text{OEE} & \text{RNH}_2 \\
 & \text{OPE} & \text{H}_2\text{O}, 25\text{°C} & \\
 & \text{OSE} & \text{R} = Ar, Alk & \\
 & 206 (60-89\%) & \\
\end{array}
$$

Introduction of a cyano group in the β -position of α , β -unsaturated trifluoromethylketones can essentially broaden their synthetic potential as building blocks. The influence of a broad range of solvents and catalysts on the reaction of CF_3 -enone 124 with TMSCN was investigated.^{[71](#page-50-0)} Depending on the solvent and the catalyst applied, individual products 207 and 208 or their mixtures can be obtained. β -Cyanoenone 209 was also prepared by treatment of 207 with con-centrated sulfuric acid.^{[72](#page-50-0)} Compound 207 in this case was synthesized by the addition of trimethylsilyl cyanide to enone 124 in the presence of a catalytic amount of iodine.^{[73](#page-50-0)}

2.4. Creation of C^1 – C^2 bonds

Acrylic acid esters can be converted into α , β -unsaturated trifluoromethylketones by addition of the Ruppert reagent (TMSCF3) to the carbonyl group of esters. Cesium fluoride was suggested as the catalyst for the addition of $TMSCF₃$ to the carbonyl group of various esters.^{[74](#page-50-0)} The intermediate acetals 210 can be hydrolyzed in acidic conditions and various ketones 211 can be obtained including enones.

$$
R^{1} = Me, Et; R = CH_{2}CMe_{3}, \quad Ph - C = C - \frac{5}{5}.
$$
\n
$$
R^{1} = Me, Et; R = CH_{2}CMe_{3}, \quad Ph - C = C - \frac{5}{5}.
$$
\n
$$
R^{2} = Me, Et; R = CH_{2}CMe_{3}, \quad Ph - C = C - \frac{5}{5}.
$$

A similar methodology was carried out for cyclohexenyl- and 4-oxocyclohexenyl-carbaldehydes 212 and 213 followed by Dess–Martin oxidation. The corresponding CF_3 -enones 214 and 215 were prepared in moderate yields. Noteworthy is the preservation of stereochemistry of the substituents in ketone 215. [75](#page-50-0)

An analogous method was applied^{[76](#page-50-0)} for the transformation of allylic alcohol 216 into the target ketone 42. In this case, Swern oxidation was used. The addition of $TMSCF₃$ to 217 gives the allylic alcohol 216 in near-quantitative yield, but the yield of target CF_3 -enone 42 was not given.

TMSCN CF3 H2SO4 (100%) OTMS EtO $\overline{C}N$ NC \diagup o $CF₃$ **124 207** (97%) 30 min, 20 °C **209** (53%)

It has been shown that a variety of aldehydes react smoothly with trimethyl(trifluoromethyl)silane in the presence of fluo-ride ions supported on an Amberlyst A-27 resin.^{[77](#page-50-0)} The reaction work up involved quenching with Amberlyst A-15 (an

acidic ion-exchange resin) followed by sequestering any unreacted aldehyde starting material using an aminomethylated (AM-polystyrene) resin. The trifluoromethyl carbinols were obtained in good yields and purity after filtration and evaporation in vacuo. In this work, various solid-supported oxidizing agents were investigated, with the best results being obtained using permanganate supported on Amberlyst A-27. The reactions were conducted in refluxing methylene chloride in the presence of 4 Å molecular sieves to act as dehydrating agent. The corresponding trifluoromethylketones were separated in good yields and purity.

Systematic studies on intramolecular Pauson–Khand reactions of various fluorine-containing enynes were de-scribed.^{[78](#page-50-0)} The CF₃-containing alkynone 218 was synthesized by formylation of acetylene 219, followed by reaction of aldehyde 220 obtained with TMSCF₃ and furnished the trifluoromethyl alcohol 221. Oxidation of alcohol 221 with Dess–Martin periodinane produced the trifluoromethylketone 218. All attempts at the direct synthesis of ketone 218 from acetylene 219 by metallation with BuLi followed by treatment with TFAA or $CF₃CO₂Et$ failed. Surprisingly, no desired products were obtained from the prepared trifluoromethylketone 218 in Pauson–Khand reaction conditions.

2.5. Creation of C^3-R^1 bonds

A new method for the preparation of α -chloro(bromo)- α , β -unsaturated trifluoromethylketones 222 has been de-scribed.^{[79](#page-50-0)} The approach is based on the addition of the appropriate halogen to the double bond of enone 124 followed by dehydrohalogenation of the intermediate dihaloketone 223. The analogous introduction of iodine was not possible, but the application of iodine chloride allowed a solution to problem. 80 The corresponding product 222 with $X=I$ can be prepared according to this procedure with a yield of 75%. In this case, the intermediate product analogous to 223 is unstable.

X2, CCl4 0 °C EtO O X CF3 X Py, CCl4 0 °C X EtO O CF3 **²²³** (~100%) **²²²** (80-85%) X = Cl, Br EtO O CF3 **124**

Halogenation of ketone 224 proceeds in two directions.^{[79](#page-50-0)} In the case of chlorination, the addition product 225 is formed, which can be transformed into 226 with base. Through brominating 224 with molecular bromine, the allylic bromination of the methyl group takes place forming the compound 227.

The chlorination reaction was performed for CF_3 -enone 228 containing a b-imidazolyl substituent, which is present as the stable nitroxyl radical. N-Chlorosuccinimide was used as the chlorinating reagent.^{[81](#page-51-0)} The α -chloroketone 229 obtained was successfully converted into the corresponding a-cyanoketone 230 by treating with potassium cyanide.

Compounds 222 similar to their non-halogenated analog 124 react with methyl- and dimethylamine forming the corresponding α -halogeno- β -amino- α , β -unsaturated trifluoromethylketones 231 , but the yields are lower in this case.^{[79,80,82](#page-50-0)} It is remarkable that, for compound 231 with R=H, the Zisomer is more stable compared with the E-isomer, which would be favorable for intramolecular hydrogen bond N– $H \cdots$ O formation. This phenomenon is obviously provoked by steric hindrance caused as a result of a trifluoromethyl group and halogen atom interaction in molecule 231 .^{[79,80](#page-50-0)} Later, by analyzing NMR and IR spectroscopy data and quantum-chemical calculation results, it was established that only the *EZE*-form of 231 is realized for α -bromo- β alkylamino- α , β -unsaturated trifluoromethylketones out of the eight forms possible (diastereomers and rotamers).[83](#page-51-0) The stability of this form is uncharacteristic for a-unsubstituted aminoketones. Two reasons for the unexpected stability are the energy factor, which favors the formation of the ap,sp.-conformer and the additional gain in energy through the possibility of weak intramolecular hydrogen bond N–H \cdots Br formation.^{[83](#page-51-0)}

The reaction of CF_3 -enaminoketones 232 with tosyl isocya-nate was investigated.^{[84,85](#page-51-0)} The reaction leads to mixture of the adducts 233 and 234. The ratio of the products depends on the substituent in the enaminoketone and mostly does not depend on the polarity of the solvent used.

The standard method for the preparation of α , β -unsaturated ketones from aliphatic ketones is treatment with phenylselenyl chloride followed by oxidation with hydrogen peroxide and elimination of PhSeOH. This method was used

for the preparation of cyclic CF_3 -enone 235 with 1,3-diketone 236 as the starting compound.^{[86](#page-51-0)} The target ketone 235 was obtained in high yield.

3. Synthesis of acetylenic CF₃-ketones

The range of methods for the preparation of acetylenic CF₃ketones is much narrow than that for the preparation of CF_{3} enones. There are only several universal methods for the preparation of acetylenic CF_3 -ketones. One of the classical methods is the trifluoroacylation reaction of anions generated from terminal alkynes under treatment with strong bases. This approach was used for the synthesis of conjugated acetylenic ketones 237 containing perfluorinated groups. Acetylenides generated from vinylacetylene 238 under treatment with n -BuLi were acylated with anhydrides of the corresponding acids.[87](#page-51-0)

$$
\begin{array}{cc}\n\hline\n\text{=} & 1 \text{ n-Buli/THF} \\
\text{OMe} & 2 \text{ (RCO)}_2\text{O} & \\
\text{238} & \text{R} = \text{CF}_3, \text{C}_2\text{F}_5 & 237 \text{ (69-90%)}\n\end{array}
$$

Trifluoroethyl trifluoroacetate can also be used for the acylation of carbanions generated from acetylenes 239. [88](#page-51-0) The target ketones 240 were obtained in yields close to quantitative. The application of TFAA, ethyl trifluoroacetate and ethyl trifluorothioacetate caused a severe decrease in the yields of 240.

An analogous reaction was used for the synthesis of trifluoroacetylacetylene 241. [89](#page-51-0) Lithium acetylenide generated from acetylene 242 was treated with ethyl trifluoroacetate in the presence of $BF_3 \cdot Et_2O$. The target ketone 241 was obtained in moderate yield and was used for further transformations in the synthesis of physiologically active compounds (histone-deacetylase inhibitors).

A second convenient method for the synthesis of acetylenic ketones containing a CF_3 group is the sequence for the preparation of secondary propargylic alcohol 243 starting with acetylene 244 and fluoral with further oxidation of 243 into ketone 245 using active manganese dioxide.^{[90](#page-51-0)} The

ketone 245 containing a CF_3 group was obtained in high yield, the value of which was not given in the article.

with atmospheric oxygen in the presence of the tetramethylammonium salt of an o-phenylene-bis-N-methyloxamidate

$$
F_3C
$$

\n
$$
F_3C
$$

\n
$$
H = CH(OEt)_2
$$

\n
$$
H_1 + H = CH(OEt)_2
$$

\n
$$
H_2H_3
$$

\n
$$
H_3C
$$

\n
$$
CH_2Cl_2
$$

\n
$$
H_3C
$$

\n
$$
CH_2Cl_2
$$

\n
$$
H_3C
$$

\n
$$
CH_2Cl_2
$$

\n
$$
H_3C
$$

\n
$$
H_3Cl_2
$$

\n
$$
CH_2Cl_2
$$

\n
$$
H_3Cl_2
$$

\n
$$
CH_2Cl_2
$$

\n
$$
H_3Cl_2
$$

\n
$$
CH_2Cl_2
$$

\n
$$
CH_2Cl_2
$$

\n
$$
CH_2Cl_2
$$

\n
$$
H_3Cl_2
$$

\n
$$
CH_2Cl_2
$$

\n
$$
H_3Cl_2
$$
<

The synthesis of acetylenic CF_3 -ketone 246 having an aryl substituent has been demonstrated. 91 The reaction of copper(I) acetylenide with trifluoroacetyl chloride was used. The acetylenide was prepared from the corresponding acetylene 247 and copper(I) bromide in an autoclave in the presence of triethylamine in toluene solution. The target acetylenic ketone 246 was isolated in good yield.

An interesting method for the synthesis of acetylenic CF_3 -ketones has been elaborated.^{[92](#page-51-0)} Esters of α -hydroxycarboxylic acids 248 containing a CF₃ group in the α -position served as the starting compounds. These compounds are obtained in good yields in the reaction of Grignard reagents 249 with ethyl trifluoropyruvate. Further hydrolysis of the esters leads to the carboxylic acids 250, which are oxidized Co(III) complex and pivalic aldehyde into the target trifluoromethylketones including acetylenic 251.

Earlier, an unknown chloromethyltrifluoroacetylacetylene 252 was prepared 93 using the following sequence: metallation of propargyl chloride 253 with n-BuLi, acylation of the lithium derivative 254 with benzyl or decyl trifluoroacetate and subsequent thermal decomposition of the acetal obtained 255 with formation of the target product.

Electrophilic substitution of a trimethylstannyl group under treatment with molecular halogens of trimethylstannyl trifluoroacetylacetylene 256 was used for the preparation of halogen derivatives of trifluoroacetylacetylenes 257. These acetylenes 256 can be prepared using the reaction of bistrimethylstannylacetylene 258 and trifluoroacetic acid anhy-dride.^{[94](#page-51-0)} An analogous approach for the synthesis of the parent trifluoroacetylacetylene 257 (X=H) was proposed.^{[95](#page-51-0)} The method is based on the reaction of trimethylstannyl trifluoroacetylacetylene 256 with trifluoroacetic acid. The target acetylene 257 (X=H) was obtained in high yield.

4. Application of α, β -unsaturated trifluoromethylketones in synthesis

The most outstanding area of application of α , β -unsaturated trifluoromethylketones is the synthesis of heterocyclic compounds, mainly due to their extensive synthetic utility, resulting in a large amount of work devoted to this problem. Heterocyclic compounds containing a trifluoromethyl group are attractive targets for medicinal chemistry and the elaboration of new effective methods for their synthesis is therefore an urgent and important task.

4.1. Heterocyclizations based on α , β -unsaturated trifluoromethylketone applications

4.1.1. Synthesis of three- and four-membered heterocycles. Numerous reactions of the perfluorinated CF_3 -enone 262 have been studied.^{[97](#page-51-0)} This ketone is formed by pyrolysis of oxolene 263, which in turn is prepared by high-temperature hydrolysis of 264 (tetrafluoroethene tetramer). The yields for this sequence are not given. The fluorinated derivatives of oxirane 265 and azetine 266 were obtained in the reaction of 262 with sodium hypochlorite and primary amines, respectively. The heterocyclic compounds were obtained in good yields.

4.1.2. Synthesis of five-membered heterocycles.

4.1.2.1. Synthesis of pyrrole derivatives. Various het-erocycles using diethoxyenone 205 were synthesized.^{[98](#page-51-0)} The reactions with several binucleophiles and sodium cyanide are described for the synthesis of the corresponding pyrrolidone 267. The reaction proceeds in good yield under reflux of the reagents in aqueous ethanol.

It was shown^{[99](#page-51-0)} that the reaction of several enones 268 with sodium cyanide gave the corresponding pyrrolidones 269 as a mixture of diastereomers. The product stereochemistry is rather complicated and was fully investigated using a combination of chromatography/mass-spectrometry and NMR spectroscopy methods.^{[99](#page-51-0)} This method seems to be more rational than that proposed earlier.^{[100](#page-51-0)} The latter method includes an additional stage of preparing the ketone monoacetal 268a, which exists in equilibrium with its hydrate. The individual diastereomers of compounds 269 were not isolated.

The same reaction was investigated for ketones 270 having no alkoxysubstituents in the β -position.^{[101](#page-51-0)} The reaction was carried out in a refluxing aqueous methanol or ethanol solution of the CF_3 -enone and sodium cyanide. The corresponding hydroxyderivatives of the pyrrolidin-2-ones 271 were obtained in good yields as a mixture of diastereomers, which were separated by column chromatography. Additionally, the dehydration reaction of the obtained products was carried out. This reaction proceeds with migration of the double bond and leads to 3-pyrroline-2-one 272 formation.

The pyrrole trifluoromethyl derivatives 273 were obtained using the reaction of ketones 274 with primary amines.^{[102](#page-51-0)} The starting heterocyclic ketones 274 were prepared from the Appel salt 79. [32](#page-50-0)

Using photolytic rearrangement of aziridine-substituted enaminoketones 275, CF₃-pyrrole derivatives were obtained.[103](#page-51-0) Depending on the substituents in the starting ketone 275, the dihydropyrrole 276 or a mixture of diphenylpyrrole 277 and dibenzoindole 278 was formed.

 β -amino CF₃-enones 281 and 282 with trimethylphosphine. The target β -alkoxy and β -amino pyrroles 284 and 283 containing a CF_3 group were obtained in good yields. β -Amino- CF_3 -enones 282 were prepared from β -alkoxyenones 281 by the reaction with secondary amines.

The viability of a reaction sequence based on the reaction of an α -amino acid 285 with the alkoxy enone 124 followed by a cyclization was established with three aminoacids with isolation of the intermediate β -dialkylamino unsaturated ketones 286.^{[106](#page-51-0)} Reaction of the proline 286a with trifluoroacetic anhydride gave the fluorinated pyrrole 287 in 70% yield. In contrast, under similar conditions, N-benzylglycine 285c gave the fluorinated pyrrolecarboxylic acid 288c in 41% yield. An obvious simplification of this pyrrole synthe-

The widely used reaction of acylation of enaminoketones 280 with oxalyl chloride was applied for the preparation of $1H$ -pyrrole-2,3-diones 279.^{[104](#page-51-0)} The reaction proceeds at room temperature and the yields of the products are not given.

A novel approach for the synthesis of pyrrole derivatives was published recently.^{[105](#page-51-0)} These heterocycles were prepared using the reaction of azidomethyl derivatives of β -alkoxy and sis might be a one-pot procedure involving the sequence of trifluoroacylation of ethyl vinyl ether, enamine formation and final cyclization of the intermediate enamine. This three-step sequence was applied to pipecolic acid 285d and thiaproline 285e. The reaction was carried out without isolation of the intermediate enaminoketones 286d,e. The corresponding pyrroles 288b, 288c, 289a, 289b, and 290 were obtained in moderate yields.

4.1.2.2. Synthesis of thiophene derivatives. Acetylenic ketone 245 was successfully used as a starting compound for the preparation of 3 -CF₃-thiophene-2-carboxylate 291. Treatment with methyl thioglycolate in THF and cesium carbonate in methanol leads to the target thiophen-2-carboxylic acid ester derivative 291 in good yield.^{[90](#page-51-0)}

4.1.2.3. Synthesis of furan derivatives. The oxidative dimerization of acetylenic ketone 292 under treatment with lead dioxide in a methylene chloride/trifluoroacetic acid mixture was carried out[.107,108](#page-51-0) The formation of the substituted furan 293 bearing CF_3 and $COCF_3$ groups was established in moderate yields.

4.1.2.4. Synthesis of pyrazoles and their derivatives.

b-Enaminoketones react with substituted hydrazines, opening up a simple and effective route to various pyrazoles. The reactions of ketone 295 with N-substituted hydrazines were carried out. The reaction leads, depending on the structure of the starting hydrazine, to individual pyrazoles or to a mixture of regioisomers 296 and 297.^{[109](#page-51-0)}

Trifluoromethylfuran derivatives 294 were also prepared by the reaction of ketone 274 with secondary amines.^{[102](#page-51-0)} The starting heterocyclic ketone 274 was prepared from the Appel salt 79. [32](#page-50-0)

was prepared from ethyl trifluoroacetoacetate and dimethylformamide dimethyl acetal (DMF-DMA). The reaction proceeds regioselectively and only one isomer was isolated in good yield.

The formation of pyrazole 300 by the reaction of enaminoketone 139 with hydrazine hydrate and a mixture of pyrazole

301 and dihydropyrazole 302 in the reaction with phenylhy-drazine was demonstrated.^{[111](#page-51-0)}

The numerous reactions of perfluorinated CF_3 -enone 262 were studied including the reaction with hydrazine.^{[97](#page-51-0)} The perfluorinated derivative, pyrazolidine 303, was obtained. This compound is a stable solid, subliming in a vacuum without decomposition. The stability of this compound can be explained by the presence of a number of fluorine atoms in the molecule.

The reaction of the primary β -ethoxy-CF₃-enone 124 with N-methylhydrazine has been re-investigated.[112,113](#page-51-0) It was demonstrated that the earlier studies contained irreproducible results and that, in fact, the reaction of 124 with N-methylhydrazine gives two isomeric dihydropyrazoles 304 and 305 in various ratios. These pyrazolines 304 and 305 undergo dehydration with different rates to form the pyrazoles 306 and 307. The yields of the products were not given.

The investigation of the above reaction was continued for β alkoxy- CF_3 -enones 308.^{[114](#page-51-0)} The formation of mixtures of pyrazole regioisomers 309, 310 and the dihydro-derivatives 311 was observed in variable yields. It was found that, using

EtO

THF or methylene chloride as the solvent, the yield of 309 containing a CF_3 group in position 3 is increased and so the selectivity of the reaction rises.

The corresponding ethoxy- and hydroxy-pyrazole derivatives 312 were obtained in good yields by the reaction of diethoxyenone 205 with hydrazine and methylhydrazine. Depending on the solvent used, the formation of the hydroxy or ethoxy derivatives 312 was observed.⁹⁸

O,N-acetals-aminals of trifluoroacetylketene 206 obtained from diethoxyenone 205 and primary amines were used for the synthesis of aminopyrazoles. The target CF_3 -containing pyrazoles 313 were prepared in good yields by the reac-tion with various hydrazines.^{[115](#page-51-0)}

The reactions of various aryl- and hetaryl-substituted hydrazines with β -ethoxy-CF₃-enone 314 containing an acetyl group in the α -position were investigated.^{[116](#page-51-0)} It was established that the heterocyclization is directed to the acetyl group for arylhydrazines and to the trifluoroacetyl group for N-methylhydrazine. The corresponding pyrazoles 315 and 316 were obtained in moderate yields.

S N

Me

N H

The synthesis of various N, N' -dimethylpyrazolium salts 317 in the reaction of enones 318 with N,N'-dimethylhydrazine dihydrochloride was reported.^{[117](#page-51-0)} The compounds 317 are very attractive, because of their potentially high herbicide activity.

The reaction of α -bromo- β -ethoxy-CF₃-enone 222 with arylhydrazines was investigated.[118](#page-51-0) The reaction was carried out by refluxing the reagents in ethanol solution, leading to the pyrazole 319 formation. The starting bromoketone 222 was prepared using bromination of ketone 124. It should be noted that the heterocyclization proceeds 100% regioselectively to open up a new effective method for the synthesis of 4-bromo-5- CF_3 -pyrazoles.

The reaction of β , β -dihalogen-substituted trifluoromethylketones 12 and 13 with N,N-dimethylhydrazine was investigated.[119](#page-51-0) The mechanism of the reaction consists of initial dimethylhydrazone 320 formation with subsequent intramolecular attack of nucleophilic fragment on β -carbon atom of vinyl group. The N,N-dimethylpyrazolium chloride 321 formed is transformed into the aromatic pyrazole 322 by nucleophilic demethylation with a second mole of dimethylhydrazine. The target 5-halogen-substituted pyrazoles 322 were isolated in moderate-to-high yields.

The same β , β -dihalogen-substituted trifluoromethylvinylketones 12 and 13 were studied in the reaction with N-ethyl-hydrazine.^{[120](#page-51-0)} The reaction proceeds analogously, but leads to the target 5-halogeno-derivatives of pyrazole 323 in higher yields.

Several pyrazoles 324 were prepared, starting from the corresponding trifluoromethylketone 325.^{[121](#page-51-0)} The cyclocondensation of 325 with hydrazine hydrochloride, methylhydrazine, and phenylhydrazine was carried out in a molar ratio of 1:1.2 using ethanol as the solvent. The use of a small excess of the hydrazine derivative was essential for good yields of the target products.

The reactions of β -ethoxy-CF₃-enones 2 with phenylhydrazine were studied. The conditions presented below were found to be best for the preparation of pyrazoles 326, 327 and pyrazoline 328.^{[122](#page-51-0)} A short reaction time and lower temperature in the reaction with pure phenylhydrazine results in the formation of pyrazoline 328, whereas reaction under elevated temperature or using the hydrochloride salt of phenylhydrazine gives the pyrazoles 326 and 327.

It was noted that the pathway of the reaction for ketone 124 with phenylhydrazine differs from the pathway of the reac-tion with the fluorinated analog.^{[123](#page-51-0)} As an example, the reaction of 124 with phenylhydrazine leads to the corresponding pyrazole 329, while the same reaction with pentafluorophe-

nylhydrazine leads to the formation of pyrazoline 330. This difference was explained by the reduced basicity of polyfluorinated arylhydrazines. The pyrazolines 330 can be dehydrated into the pyrazoles 331 using phosphorus pentoxide in chloroform.

An interesting example of the application of trifluoroacetylpyrroline 20 for the preparation of pyrazoles 332 has been described.[124](#page-51-0) The starting enaminoketone 20 can easily be obtained by trifluoroacylation of the N-substituted proline 19 with TFAA. It was found that this compound is a new 1,3-ambidentate electrophile reacting with bifunctional Nnucleophiles such as hydrazines and amidines to give CF_3 substituted pyrazoles bearing a β -aminoethyl side chain. The reaction sequence represents a special type of ring transformation by ring-chain transfer, where a ring and a chain

moiety in the adduct are transformed into each other, giving the product. In view of the pharmacological interest in heterocycles bearing both a CF_3 appendage and a β -aminoethyl side chain, the method is very attractive.

The reaction of 4-hydrazo-7-chloroquinoline 340 with various β -methoxy-CF₃-enones 341 was investigated.^{[126](#page-51-0)} The aim of the research was to generate new compounds for further antimalarial screening. The reaction was performed in

The reaction of hydrazine with cyclic CF_3 -enones— β -trifluoroacetyldihydropyran 145 and β -trifluoroacetyldihydrofuran 333—was studied.[88](#page-51-0) The reaction was carried out in ethanol solution. It was demonstrated that, in the case of b-trifluoroacetyldihydropyran 145, the reaction leads to the corresponding pyrazole 334, while, in the case of β -trifluoroacetyldihydrofuran 333, the formation of 335 at the first stage is observed. Monomeric pyrazole 336 can be prepared by treatment of an ethanol solution with hydrogen chloride.

The reactions with various binucleophiles including the reactions with hydrazines were investigated for the CF₃enones 337 and 338 containing a dialkyldithio fragment in the β -position (ketene dithioacetals) and so the possibility for the preparation of pyrazoles 339 or containing a 1,3-di-thiopropyl substituent was shown.^{[125](#page-51-0)}

methanol at reflux. Depending on the substituent in the starting CF_3 -enone, the dihydropyrazoles 342 were obtained in good yields. Dehydration into the corresponding pyrazoles 343 was performed under reflux in acetic acid. In some cases, simultaneous dehydration was observed at the stage of reflux in methanol. The same reaction of various β -methoxy- β aryl-substituted CF_3 -enones 341 with hydrazoquinoline 340 was applied for the preparation of several CF_3 -containing pyrazoles 343 possessing high antimalarial activity.[127](#page-51-0) The reaction proceeds with the formation of the intermediate pyrazoline 342 . The starting CF₃-enones 341 are available through trifluoroacylation of the acetophenone enol ethers.

Using a twofold excess of the ketones 344 in the reaction with aminoguanidine hydrocarbonate, the formation of pyrazolinylpyrimidines 345 is observed. These compounds can be easily dehydrated into the corresponding pyrazolylpyr-imidines 346.^{[128](#page-51-0)} This transformation allows the simultaneous creation of two heterocyclic systems in one step. In addition, many of the 2-pyrazolylpyrimidine derivatives show a high physiological activity.

The reaction of enones 347 with a 2-hydrazo derivative of pyrimidine 348 proceeds in an analogous manner. The corresponding pyrimidine derivatives 349 containing a dihydropyrazole substituent in position 2 of the pyrimidine ring are formed.^{[129](#page-51-0)} These products are potential analgesics and antipyretics.

An interesting example is the reaction of ketone 71 with phenylhydrazine. The tetrahydropyrazole derivative 350 was obtained in moderate yield. The reaction proceeds with 100% stereoselectivity—only the diastereomer of 350 shown was obtained.^{[130](#page-51-0)}

The reaction of β -alkoxy- β -aryl-CF₃-enones 351 with thiosemicarbazide as the hydrazine derivative 352 was studied.[131](#page-51-0) The products of the reaction are the corresponding hydroxyl derivatives of dihydropyrazoles 353. These compounds were obtained in high yields. They can be transformed into the N-substituted pyrazoles 354 in high yields using acidic hydrolysis.

The ketone 227 was obtained by allylic bromination with molecular bromine. Having a hidden bromoketone fragment in the structure, this compound was used for heterocylization using the reaction with 355, forming a thiazole connected with pyrazoline 356.^{[132](#page-51-0)}

The reaction of ketones 357 ($R_2=H$) with 2-pyridylcarboxamidrazone 358 leads to the predominant formation of the pyrazoline derivatives 359 , 133 regardless of the fact that amidrazones possess three reactive nitrogen atoms. The reaction is accompanied with imine fragment of amidrazone hydrolysis. The hydrochlorides 359 can be transferred into the free bases using triethylamine/diethyl ether solution. Attempts at dehydration of the free bases into the corresponding pyrazoles were unsuccessful.

The preparation of copper(II) amidrazone complexes 360 derived from CF_3 -containing β -alkoxyenones 361 was described recently.^{[134](#page-52-0)} Ketones 361 react with 2-pyridylcarboxamidrazone 362 to produce the corresponding 1,1,1-trifluoro-4-aryl-4-(N-pyridine-2-carboxamidrazone)-3-buten-2 ones 363. The compounds 363 react with copper(II) chloride to give 1:1 adducts, in which the donor molecules were shown to isomerize to their cyclic pyrazolic forms. The coordination chemistry of the products derived from this

 $R = Me$, Et; R₁ = H, Me, Ph, 4-MeC₆H₄, 4-BrC₆H₄, 4-NO₂C₆H₄; R₂ = H, Me

reaction with copper(II) chloride was investigated. The adducts 360 containing the amidrazone pharmacophore have been tested as anticancer drugs.

The reaction of acetylenic CF_3 -ketone 246 with hydrazine hydrate was used for the preparation of the CF_3 -substituted pyrazole 364.^{[91](#page-51-0)} The reaction was performed in toluene under reflux. The corresponding pyrazole 364 was obtained in excellent yield.

An analogous reaction was used for the preparation of pyrazoles 365 having aryl substituents on the nitrogen atom.^{[135,136](#page-52-0)} Acetylenic CF₃-ketones 366 react with N-arylhydrazines 367 under reflux in ethanol to give the pyrazoles 365. The yields of 365 are not given.

Work reported by our group in 2003^{137} 2003^{137} 2003^{137} is devoted to $[2+3]$ dipolar cycloaddition of various diazo compounds 368 to α, β -unsaturated trifluoromethylketones 369. The ketones 369 react with diazoalkanes 368 forming pyrazolines 370. The reaction proceeds 100% regioselectively and with high stereoselectivity. In the case of phenyldiazomethane, the exclusive formation of the cis-isomers of pyrazolines 370 is observed. Ethyl diazoacetate 371 gives the cis-isomer predominantly $(\sim 75\%$ of the mixture). The ketone 369 having R^1 =Ph and R^2 =Me does not react with aryldiazoalkanes, although it forms the cycloaddition product in the case of reaction with diazomethane (regiochemistry of the product is analogous to 370). Using the trifluoroacetylated acetylene 372 in the reaction with ethyl diazoacetate 371 allows the preparation of the pyrazole 373 .^{[137](#page-52-0)}

³⁷³ (45%) **³⁷² ³⁷¹**

4.1.2.5. Synthesis of isoxazoles and their derivatives. O-Vinyl oximes 6 react readily with trifluoroacetic anhydride in the presence of pyridine at room temperature to give CF_3 enones. When the reaction mixture was treated, after trifluoroacylation, with aqueous Na HCO_3 , 4,5-dihydro-1,2-oxazole 374 was isolated as the single product.^{[138](#page-52-0)} The formation of 374 implies the hydrolysis of 7 via an intermediate semiacetal-like adduct 375, which decomposes to 3-oxo-4,4,4-trifluorobutyraldehyde 376 and acetoxime 377. The two latter compounds undergo re-oximation to result in the corresponding aldoxime 378 and acetone 379—the hydroxylamine exchange between oximes and aldehydes or ketones under solvolytic conditions is a well-established fact.

The isoxazoline 380 was synthesized starting from the CF₃enone 325. [121](#page-51-0) The presence of a methoxycarbonyl group in the isoxazole obtained makes possible further transformations.

325 MeOH, reflux **380** (70%)

NH2OH·HCl

O $F_3C \rightarrow \heartsuit$ OMe

N H

N

 F_3C

 M e O_2C

The reaction of a number of β -methoxy- β -aryl-CF₃-enones 381 with hydroxylamine hydrochloride was investigated.^{[139](#page-52-0)} The formation of 4,5-dihydroisoxazoles 382 was established. These compounds were obtained in high yields and can be transformed into the corresponding isoxazoles 383 in yields close to quantitative using concentrated sulfuric acid. In addition, the compounds 383 can be directly obtained using the reaction of β -methoxy- β -aryl-CF₃-enones 381 with hydroxylamine hydrochloride and an excess of hydrochloric acid.

Analogous reactions were carried out for β -methoxy-CF₃enones 2 containing 2-thienyl or 2-furyl substituents. The reaction with hydroxylamine hydrochloride leads to the di-hydroisoxazole derivatives 384.^{[122](#page-51-0)} Similarly, the products obtained can be easily dehydrated with concentrated sulfuric acid, forming the corresponding isoxazoles 385.

The reaction of diethoxyenone 205 with hydroxylamine hydrochloride has been investigated. The corresponding ethoxy derivative of isoxazoline 395 was synthesized in good yield[.98](#page-51-0) Similarly, O,N-acetals-aminals of trifluoroacetylketene 206 were used for the synthesis of amino-substituted isoxazoles.^{[115](#page-51-0)} The target CF_3 -containing isoxazolines 396 were prepared in good yields.

An attractive sequence of reactions was carried out for a cyclic β -alkoxy-CF₃-enone 386.^{[140](#page-52-0)} This allows the preparation of isoxazoles 387 and 388 and their dihydro-derivatives 389 and 390 containing functional groups. These compounds were obtained in high yields. The starting 3-trifluoroacetyl dihydropyran 386 can be easily prepared by trifluoroacylation of methoxydihydropyran 391 with trifluoroacetic anhydride in the presence of pyridine.

[2+3] Cycloaddition of β -ethoxy-CF₃-enone 124 with Nmethyl-C-aryl nitrones 397 results in the isoxazolidines 398. These compounds cannot be isolated in a pure form, due to transformation to the diol 399 and the ethanol elimination product 400 under column chromatography purification.^{[142](#page-52-0)}

Ketones containing no alkoxy groups in the β -position 401 can also be used for the preparation of isoxazoles 402. The

The reaction of β -methoxy-CF₃-enones 392 with N-methyl-hydroxylamine hydrochloride was investigated.^{[141](#page-52-0)} The reaction was carried out in methanol in the presence of potassium carbonate. Depending on the substituent in the starting ketone 392, the reaction proceeds as a Michael addition, forming 393, or successive heterocyclization takes place, with regioselective formation of the isoxazolines 394.

diaryl-substituted isoxazole 402 was synthesized using the reaction with hydroxylamine with further aromatization by treatment with iodine.^{[143](#page-52-0)} An unusual stereochemistry for the heterocyclization reaction is observed.

4.1.2.6. Synthesis of oxazoles. The reaction of acetylenic CF3-ketones 245 and 403 with methyl isocyanoacetate

 $b = KI, I_2, \text{NaHCO}_3, H_2O-THF, \text{reflux}, 7 h, rt, 12 h (88%)$

404 catalyzed with silver perchlorate was investigated. The reaction leads to the formation of the dihydrooxazole deriv-atives 405.^{[144,145](#page-52-0)} It was found that the most effective catalyst in this reaction is a silver perchlorate/triethylamine system. It was shown that, in the case of CF_3 -ketones, the reaction proceeds 100% stereoselectively. The target products are formed in high yields.

2,3-dihydro-5-benzylamino-4-trifluoroacetylfurane 409 as the minor component of an inseparable product mixture.

$$
\text{CNCH}_{2}CO_{2}Me + \n\begin{array}{c}\nO \\
\hline\n\end{array}\n\right\} = R_{1} \n\begin{array}{c}\nAgClO_{4} (2\%)/Et_{3}N \\
\hline\nSO_{2} 45 R_{1} = Ph \n\end{array}\n\quad\nR_{1} \n\begin{array}{c}\nF_{3}C \\
\hline\n\end{array}\n\right\} + R_{1}CO_{2}Me\nR_{1} = Ph (96\%) \\
O\n\begin{array}{c}\n0 \\
\hline\n\end{array}\n\right\} + \nR_{2}CO_{2}Me\nR_{1} = Ph (96\%) \\
403 n\text{-Hex}
$$

4.1.2.7. Synthesis of isoselenazoles. An approach to the synthesis of scarcely available heterocycles—isoselenazoles containing a trifluoromethyl group—has been described.^{[69](#page-50-0)} Isoselenazoles 406 can be easily prepared by successive treatment of enones 203 with bromine and ammonia. In the first stage, the compounds 203 react with bromine in dichloromethane at -70° C and, in a second step, at the same temperature, an excess of gaseous ammonia is bubbled into the mixture.

4.1.2.8. Synthesis of triazoles. The reaction of ketone 124 with various azides leads to the formation of the corre-sponding trifluoroacetyltriazoles 407.^{[146](#page-52-0)} The ketones 407 obtained are easily hydrated at the carbonyl group, forming the diols 408. Attempts to carry out the [2+3] cycloaddition for 2,3-dihydro-4-trifluoroacetylfuran 333 with benzyl azide led, instead of the expected triazole, to the formation of the

4.1.3. Synthesis of six-membered heterocycles.

4.1.3.1. Synthesis of pyridines and their derivatives. Pyridines are a very important class of heterocyclic compounds and there are plenty of methods for their synthesis. There are, however, relatively few methods for the preparation of CF_3 -containing pyridines. A method for the synthesis of 6-CF₃-nicotinonitrile 410, based on the reaction of β -ethoxy CF_3 -enone 124 with β -dimethylaminoacrylonitrile 411 followed by treatment of the intermediate product 412 with ammonium acetate, was proposed.^{[147](#page-52-0)} This sequence represents a convenient method for the regioselective preparation of 6 -CF₃-nicotinonitrile 410.

Recently, a method for the synthesis of 2-arylamino-6- CF_3 derivatives of nicotinonitrile 413 was elaborated.^{[148](#page-52-0)} β -*i*-Butoxy CF_3 -enone 414 was employed as the CF_3 -containing building block. The key step of the method is the reaction of the CF₃-enone 414 with β , β -diamino-substituted acrylonitriles 415 generated in situ in the reaction of 416 with anilines. The nicotinonitriles 413 containing various arylamino substituents in the α -position of the pyridine ring were obtained in low-to-high yields.

A novel method for the synthesis of CF_3 -pyridines based on the reaction of enone 124 with N-acylacetamidrazones 417 was developed.^{[149](#page-52-0)} The reaction proceeds in good yields and allows the preparation of various 2-hydrazo derivatives of ethyl 6-trifluoromethylnicotinate and 6-trifluoromethylnicotinonitrile 418.

enaminoketone 425 with ammonium acetate in dimethylformamide.

The same method was used for the preparation of 4-arylamino-2- CF_{3} -pyridine derivatives 426 possessing anticancer activity.^{[152,153](#page-52-0)} 2-Aminobenzoic acid $\widehat{427}$ and 2-aminonicotinic acid methyl ester 428 were used as the aromatic amines. The intermediate enaminoketones 429, 430, and the dienones 431, as well as the target pyridines 426, were prepared in high yields.

A Gantsch-type synthesis of 1,4-dihydropyridine derivatives has been suggested.^{[154](#page-52-0)} This method uses the reaction of dihydrothiophene-3(2H)-one-1,1-dioxide 432 with CF_3 -enone 433. The intermediate compounds 434 were isolated as

A new approach to the synthesis of 2 -CF₃ pyridines 419 containing various arylamino substituents in the 4-position of the pyridine ring was proposed.[150](#page-52-0) This method exploits the reaction of β -methoxy CF₃-enone 224 with various aromatic amines including heterocyclic amines. The subsequent reaction of the formed enaminoketone 420 with the DMF dimethyl acetal leads to dienones 421. The final stage is the reaction of the obtained dienones 421 with ammonium acetate in dimethylformamide, leading to the target 2 -CF₃-4arylaminopyridines 419 in high yields.

a mixture of diastereomers and, without further purification, were utilized in the next reaction. The target 1,4-dihydropyridine derivative 435 was prepared in good yield.

4-Amino-2- CF_3 -pyridine 436 was prepared in moderate yield using the reaction of ketone 237 with ammonia under heating at high pressure.^{[87](#page-51-0)}

Several publications are devoted to methods for the synthesis of pyridine derivatives using lithium azaenolates. The

An analogous approach based on the use of 2-aminopyridine derivatives 422 as aromatic amines for the synthesis of the 2- CF_3 -4-pyridylaminopyridine derivatives 423 was suggested.^{[151](#page-52-0)} The target products 423 are formed in high yields by reflux of the dienone 424 obtained from

application of ketone 205 for the synthesis of substituted 4-ethoxy-2-trifluoromethylpyridines 437 has been described. The ketone 205 was involved in the reaction with azaenolates 438 prepared from lithiated alkyltrimethylsilanes 439 and aromatic nitriles.[155](#page-52-0)

Besides the ketones 443, some other α , β -unsaturated ketones react similarly in these conditions. In one example, the ketone 42 was used for the synthesis of the trifluoromethylpyridine 444 by the reaction of 42 with lithiated imine 445.^{[157](#page-52-0)} The intermediate dihydropyridine derivative transforms spontaneously into the target pyridine.

This method was extended to the preparation of various isoxazolyl-substituted pyridines 440. ^{[156](#page-52-0)} The method is based on metallation of 3-methyl-5-trimethylsilylmethylisoxazole 441. The reaction with benzonitrile gives the azaenolate 442. Subsequent treatment with CF_3 - β -alkoxyenones 443 results in the formation of the target pyridines 440 in low-to-high yields.

Enaminoketone 446 was studied as a building block for the preparation of 4-CF₃-containing pyridines 447.^{[158](#page-52-0)} Ketone 446 reacts easily with various 1,3-dicarbonyl compounds 448 in the presence of trifluoroacetic acid under mild conditions to give the α -trifluoromethylpyridines 447 in moderate-to-high yields. In several cases, the formation of a mixture of regioisomers 447 was observed. This synthetic method provides a facile and convenient access to pyridines having a trifluoromethyl group at the α -position, which are not easily obtained by other methods. The peculiarities of the reaction regiochemistry are not discussed.

hydride, the formation of the pyridine derivative 457 was observed in moderate yield.

The reaction of CF_3 -enone 224 with cyanothioacetamide was carried out.^{[163](#page-52-0)} Depending on the reaction conditions,

The same method for the preparation of nicotinates 449 involving the reaction of β -enaminoketone 446 with 1,3-ketoesters 450 was described.^{[159](#page-52-0)}

A 2-CF₃-pyridine derivative 451 was prepared in very low yield by the reaction of the enamine 452 with ammonium acetate and the ketone 42 by refluxing in triglyme.^{[160](#page-52-0)}

The synthesis of 2-hydroxy-3-nitro-6-trifluoromethylpyridine 453 starting from CF₃-enone 454 and nitroacetamide 455 has been described.^{[161](#page-52-0)} The yield for the pyridine 453 was not given.

The reaction of enone 124 with acetoacetamide 456 was studied.[162](#page-52-0) In the presence of catalytic amounts of sodium the isomeric trifluoromethyl-substituted pyridinethiones 458 and 459 were prepared in good yields.

A similar method has been reported for the preparation of pyridine-2-thiols as their N-methylmorpholine salts 460 using the reaction of enones 461 and cyanothioacetamide in the presence of twofold excess of N-methylmorpholine.^{[164](#page-52-0)} The salts 460 can be involved in further transformations, forming thienopyridines in particular.

$$
\begin{array}{cccc}\n & R \\
 & \searrow & \circ & \circ & \circ \\
 & \searrow & \circ & \circ & \circ \\
 & \searrow & \circ & \circ & \circ \\
 & \searrow & \circ & \circ & \circ \\
 & \searrow & \circ & \circ & \circ \\
 & \searrow & \circ & \circ & \circ \\
 & \searrow & \circ & \circ & \circ & \circ \\
 & \searrow & \circ & \circ & \circ & \circ \\
 & \searrow & \circ & \circ & \circ & \circ \\
 & \searrow & \circ & \circ & \circ & \circ \\
 & \searrow & \circ & \circ & \circ & \circ & \circ \\
 & \searrow & \circ & \circ & \circ & \circ & \circ \\
 & \searrow & \circ & \circ & \circ & \circ & \circ \\
 & \searrow & \circ & \circ & \circ & \circ & \circ \\
 & \searrow & \circ & \circ & \circ & \circ & \circ \\
 & \searrow & \circ & \circ & \circ & \circ & \circ & \circ \\
 & \searrow & \circ & \circ & \circ & \circ & \circ & \circ \\
 & \searrow & \circ & \circ & \circ & \circ & \circ & \circ \\
 & \searrow & \circ & \circ & \circ & \circ & \circ & \circ \\
 & \searrow & \circ & \circ & \circ & \circ & \circ & \circ \\
 & \searrow & \circ & \circ & \circ & \circ & \circ & \circ \\
 & \searrow & \circ & \circ & \circ & \circ & \circ & \circ \\
 & \searrow & \circ & \circ & \circ & \circ & \circ & \circ \\
 & \searrow & \circ & \circ & \circ & \circ & \circ & \circ \\
 & \searrow & \circ & \circ & \circ & \circ & \circ & \circ \\
 & \searrow & \circ & \circ & \circ & \circ & \circ & \circ \\
 & \searrow & \circ & \circ & \circ & \circ & \circ & \circ \\
 & \searrow & \circ & \circ & \
$$

A novel method for the preparation of CF_3 -containing pyr-idines 462 was elaborated recently.^{[165](#page-52-0)} The sequence includes several reactions. The first step is the synthesis of α -hydroxydihydropyrans 463 by the reaction of α , β -unsaturated ketones 270 and α -cyanoacetophenones 466. The second step is transformation of 463 with ammonium acetate in ethanol to form the tetrahydropyridines 464. The penultimate step is the dehydration of tetrahydropyridines 464 to give the dihydropyridines 465. The final stage is oxidation into the target pyridines 462 with DDQ. All compounds were prepared in good yields. This sequence is a very effective and useful method for the preparation of CF_3 -containing nicotinonitriles 462.

Original methods for the preparation of $4-CF_3$ -nicotinic acid 467 ,^{[166](#page-52-0)} 4-CF₃-nicotinic acid esters 468^{167} 468^{167} 468^{167} and

4-CF₃-nicotinonitrile 469 were patented recently.^{[168](#page-52-0)} The synthetic sequence consists of the reaction of the parent enaminoketone 446 with B-substituted acrylic ester 470 or b-substituted acrylonitriles 471 having a leaving group in the b-position. Under treatment with sodium hydride in DMF these compounds produce enamines 472a and 472b after elimination of the leaving group. Compounds 472a,b gave 4-CF3-nicotinic acid ester 468 or nitrile 469 in excellent yields under treatment with sodium methoxide in methanol. If water is added to the reaction mixture after cyclization, 4-CF3-nicotinic acid 467 is obtained.

An alternative one-pot synthesis of 4-trifluoromethyl-2(1H) pyridone 473 using β -ethoxy-CF₃-enone 124 and chloroace-tonitrile has been demonstrated.^{[170,171](#page-52-0)} It was found that chloroacetonitrile reacts with 124 in tetrahydrofuran as the solvent in the presence of zinc powder and trimethylchlorosilane to produce the β -trimethylsilyloxynitrile 475 accompanied by formation of the elimination product 476. The mixture of 475 and 476 was heated under reflux in concentrated HCl to give 4-trifluoromethyl-2-pyridone 473 in good yield. It was suggested that the mechanism of heterocyclization involves a Reformatsky reaction to give the inter-

4-Trifluoromethyl-2(1H)-pyridone 473 has been synthesized recently.^{[169](#page-52-0)} The reaction sequence includes olefination of ketone 124 leading to a mixture of two products 474 in high total yield and the subsequent reaction with formamide or ammonium formate, leading to the target pyridone 473.

mediate 475, which undergoes elimination of TMSOH to give 476 and further cyclization to afford 473. Chlorination of the pyridone 473 using POCl₃ resulted in the desired 2-chloro-4-trifluoromethylpyridine 477 in good yield.

Treatment of the enones 124 or 454 with 8 equiv of magnesium and chlorotrimethylsilane in DMF leads to formation of difluoro derivative of Danishefsky diene 478, which can be used in Diels–Alder reactions with various dienophiles.[172](#page-52-0) In particular, various imines 479 can be used as dienophiles. The scarcely available 5,5-difluoro derivatives of dihydropyridone 480 were obtained in good yields by this method.

4.1.3.2. Synthesis of pyrans and their derivatives. Besides the above-mentioned work, Diels–Alder reaction of difluorinated Danishefsky-diene analog 478 was used in the reaction with various aldehydes. The corresponding pyran-4-ones 481 were obtained in moderate vields.^{[172](#page-52-0)} In the same work, the asymmetric synthesis of dihydropyrone 482 using a Ti (IV)-(R)-BINOL system is described. In the reaction with benzaldehyde, a sufficiently high ee was demonstrated for the prepared pyran 482.

investigated.[142](#page-52-0) The reaction leads to unexpected products—the cycloadducts 485 having the alkoxy group migrated. The formation of a cis-/trans-isomeric mixture of 485 is observed. The corresponding dihydropyrans 485 containing a trifluoroacetyl group were isolated in moderate yields.

The influence of various Lewis-acid additives on the cycloaddition reaction of β -alkoxy CF₃-enones 124 and 486 with

The ketone form of compounds 169 can be used as a heterodiene in the Diels–Alder reaction. This method has been used in the approach to dihydropyrans 483 .^{[173](#page-52-0)} The diol form of 169 can also be used in this reaction, because it exists in the equilibrium with its ketone form. The reaction at room temperature leads to the mixture of regio- and stereoisomeric products. As an example, the reaction with phenyl vinyl sulfide leads to a mixture of cis- and trans-isomers. The application of compound 169 having R=Ph leads to resinifi-cation of the reaction mixture.^{[173](#page-52-0)}

The cycloaddition reactions of α , β -unsaturated aldehydes 484 with β -alkoxy-CF₃-enones 124 and 414 were

vinyl ethers 15 and 487 was investigated.^{[174](#page-52-0)} The best results—the highest ratio of product diastereoisomers 488-were obtained using titanium(IV) chloride.

The preparation of chiral CF_3 -dihydropyrans 490 was investigated. In this case, the reaction of CF_3 -enone 489 containing a chiral substituent in the β -position was used.^{[175](#page-52-0)} The

application of titanium(IV) chloride as the catalyst permits the preparation of the target pyrans 490 in high yields. The diastereoselectivity of the reaction is, however, very low.

Solid-phase methodology can be successfully applied to the cycloaddition reaction of β -benzyloxy-CF₃-enone 486 and vinyl ether 491. The reaction is catalyzed with a europium(III) complex and proceeds in rather moderate yield, although with high stereoselectivity. The target pyran 492 was obtained from 493 after treatment with lithium triethylborohydride.[174](#page-52-0)

OBn

486

O

 F_3C^2 \bigcirc

 \circ \sim \circ

Cyano-substituted dihydropyrans 494 were obtained in the reaction of CF_3 -enones 270 with aromatic α -cyanoketones $495.$ ^{[176](#page-52-0)} The reaction proceeds in *i*-propanol in the presence of calcinated potassium fluoride as the base. The reaction proceeds in good yields and 100% stereoselectively. The only diastereomer of the target dihydropyran 494 obtained had an equatorial orientation of the aryl substituent and the $CF₃$ group.

Pyran derivatives were also obtained in the reaction of CF_3 enone 42 with 4-methylthiophenol 496. [177](#page-52-0) A mixture of the Michael adduct 497 and the cyclic product of double addition 498 was formed. Depending on the reaction conditions, each of the two products can be obtained selectively. Although 32 isomers of 498 are possible to form in this reaction, the compound 498 was obtained as the single diastereomer and its structure was confirmed by X-ray analysis. Even the configuration of the carbon atom outside the pyran ring is fixed.

ring. These compounds were obtained as a single diastereomer, the structure of which was confirmed using X-ray analysis. The trifluoromethyl group and the benzothiazoline substituent are oriented equatorially.

The investigation of reactions of β -ethoxyenones 500 with various N-aroyl glycines 501 in the presence of acetic acid anhydride was studied.^{[179](#page-52-0)} The heterocyclization proceeds upon heating and leads to 2H-pyran-2-one derivatives 502 in high yields.

4.1.3.3. Synthesis of pyrimidines and their derivatives. Bonacorso et al. reported the synthesis of pyrimidones 504 using the reaction of enones 503 with urea in acidic conditions.^{[180](#page-52-0)} In the case of ketones having R=Me, the yield of the pyrimidines 504 decreases and the reaction time increases. Apparently, this is due to steric hindrance caused by the presence of a methyl group in the α -position.

Enamidoketone 73 was used for the preparation of pyrimidine derivative 505. [30](#page-50-0) The heterocyclization was carried out under basic conditions. The yield of the product 505 was not given.

The reaction of 2-aminothiopenol 126 with two cyclic β alkoxyenones 145 and 333 was investigated.¹⁷⁸ Using toluene as a solvent, the reaction leads to the formation of benzothiazolines 499 bound to a tetrahydrofuran or a tetrahydropyran The reaction of a series of β -alkoxy CF₃-enones **506** with acet- and benzamidine 507 was carried out.^{[181](#page-52-0)} The formation of pyrimidines 508 or the mixture of 509 and their tetrahydro derivatives 509 is observed. The ratio of 508 and

509 varies within wide limits, depending on the reaction conditions and the reaction time, as well as on the substituents in the starting compounds. The individual compounds 508, 509 or their mixtures were obtained. The stereochemistry of the tetrahydropyrimidine products 509 has not been investigated.

preparation of 510 in high yields.^{[182](#page-52-0)} The amidines 511 were generated in situ from the hydrochlorides by treating with sodium hydroxide.^{[182](#page-52-0)}

The reactions of CF_3 -enone 71 containing an ethoxycarbonyl group in the α -position with thiourea and guanidine sulfate were investigated.^{[130](#page-51-0)} The corresponding dihydro 513 and tetrahydro derivatives 514 were obtained in moderate yields. The stereochemistry of 514 was not studied.

In order to study the regiochemistry of the cyclocondensation reaction of a series of β -alkoxyvinyl CF₃-ketones 515 toward an asymmetric dinucleophile, N-methylthiourea was chosen.^{[183](#page-52-0)} The cyclocondensation was carried out in methanol under acid catalysis. Depending on the temperature and the reaction time, the open-chain products 516 or pyridinethiones 517 were obtained. In general, a low temperature and a short reaction time promote the formation of the open-chain products 516. Higher temperatures and longer reaction times results in the pyrimidinethiones 517. The open-chain products 516 were isolated for the first time. These compounds are difficult to

The synthesis of trifluoromethyl-substituted pyrimidines 510 was described using the reaction of enones 270 with amidines 511. Subsequent dehydration and oxidation of the intermediate adducts 512 without isolation permits the

isolate because the cyclization usually takes place very rapidly. The compounds were isolated probably due to the steric hindrance between the N -methyl group and the $CF₃$ group.

Some work 184 has been undertaken to apply the above methodology to the synthesis of fluorinated aminopyrimidines analogous to trimethoprim (TMP). TMP and pyrimethamine (PYR) have become the reference drugs for the prophylaxis and treatment of opportunistic infections due to Pneumocystis carinii and Toxoplasma gondii. Enaminoketones 518 were reacted with guanidine hydrochloride to give the pyrimidines 519. Anti-Toxoplasma activities of the derivatives 519 were assessed in vitro using a tissue-culture model combined with an immunoenzymatic assay for quantification of Toxoplasma growth and TMP was tested in parallel as a reference drug.

Four novel pyrimidines were prepared to investigate their effects on NTPDase activity in a synaptosomal fraction obtained from rat cerebral cortex[.185](#page-52-0) The dihydropyrimidine 520 was prepared by the cyclocondensation reaction of 224 with 1,2-dimethyl-isothiourea sulfate in the presence of sodium hydroxide. The synthesis of 521 was achieved from the cyclization of the pyrimidin-2-yl-hydrazine 522 with the ketone 224 in chloroform under reflux for 24 h. The pyrimidine 523, prepared by the oxidation of 2-methylsulfanyl-pyrimidine 524 with MCPBA, underwent nucleophilic displacement of the 2-methylsulfonyl group by hydrazine hydrate in ethanol and reflux to furnish the 2-hydrazinopyrimidine 522 in excellent yield. In general, all the novel pyrimidines tested as inhibitors of NTPDase showed a similar inhibition for both substrates ATP and ADP. It should be noted that the presence of the pyrimidine core was essential for enzyme recognizing of the compounds in both peripheral and active sites.

Recently, much attention has been devoted to the preparation and subsequent functionalization of chlorinated and brominated (trifluoromethyl)pyrazoles, (trifluoromethyl)pyri-dines, and (trifluoromethyl)quinolines.^{[187](#page-53-0)} An extension of this work to the pyrimidine analogs carrying a trifluoromethyl group and an additional halogen as substituents was reported. The condensation of urea with ketone 124 afforded $4-(\text{trifluorometry})$ pyrimidin-2(1H)-one 531, a known compound, in good yield. This was converted into 2-bromo-4-(trifluoromethyl)pyrimidine 532 by reaction with phosphorus tribromide. The analogous condensation of urea with 3-bromo-4-ethoxy-1,1,1-trifluoro-3-buten-2-one 222 (X=Br) produced tars, rather than the expected 5-bromo-4-(trifluoromethyl)-2(1H)-pyrimidone.

The bromoketone 227 has been applied for various heterocyclizations.[79](#page-50-0) Compound 227 reacts with methylisothiouronium sulfate forming the corresponding pyrimidine derivative 533 in moderate yield.

A library of pyrimidine derivatives including compounds 525 and 526 containing a CF_3 group was synthesized for an investigation of their physiological activity.[186](#page-52-0) Two main approaches for the synthesis were used. The first approach was based on the reaction of β -ethoxy-CF₃enone 527, which is a derivative of trifluoroacetoacetic ester, with urea or amidines. The second approach was used for preparation of 526 is based on the reaction of formylenaminoester 528 obtained by formylation of malonic acid monoester potassium salt 529 with trifluoromethylamidine 530.

The reactions of 2-guanidinopyrimidine 534 with β -alkoxy- CF_3 -enones 535 and cyclic enones 145 and 333 were investigated.[188](#page-53-0) Depending on the substituents in the starting ketone, the reaction leads to dipyrimidylamines 536 or their condensed dihydrofuran and dihydropyran derivatives 537.

An analogous reaction was used for the synthesis of a 2-dimethylamino derivative of 4 -CF₃-pyrimidine-5-carboxylic acid 538 showing cardiotonic activity.^{[189](#page-53-0)} The reaction of β -ethoxy CF₃-enone 72 with 2,2-dimethylguanidine was performed. The target product 538 was obtained in high yield.

The possibility of application of β -enamino-CF₃-enone 539 for the synthesis of pyrimidine derivatives was shown.^{[190](#page-53-0)} The pyrimidines 540 were obtained in good yields using the reaction with N,N-dimethylguanidine, guanidine, and O-methylurea.

It was found that β , β -bis(trifluoroacetyl)enamine 541 can be easily prepared in two steps with trifluoroacetic anhydride and isobutyl vinyl ether with a subsequent i -BuO–NH₂ exchange reaction with ammonia.^{[191](#page-53-0)} Enaminoketone 541 reacts with various aldehydes in the presence of ammonia to give the dihydropyrimidine derivatives in good yields. Oxidation of 1,2-dihydropyrimidines 542 with DDQ at room temperature for 24 h in acetonitrile caused smooth dehydrogenation to give the desired pyrimidines 543. This synthetic method provides a facile and convenient access to pyrimidines 543, having both trifluoromethyl and trifluoroacetyl groups, which are not easily obtained by other methods.

 $R = H$, Me, Et, *i*-Pr, *p*-MeOC₆H₄, *p*-MeC₆H₄, Ph, *p*-ClC₆H₄

Cyclic β -alkoxy-CF₃-enones **544** were applied for the preparation of 2-pyrimidones 545 and their thio analogs using the reaction with urea and thiourea.[192](#page-53-0) The target compounds 545 were obtained in moderate yields.

The reactions of CF_3 -enone 337 with several N,N-binucleophiles were investigated. Various 2-substituted pyrimidines 546 containing a 1,3-dithiopropyl substituent were pre-pared.^{[125](#page-51-0)}

An example of the application of trifluoroacetylpyrroline 20 for preparation of pyrimidines 547 was also described.^{[124](#page-51-0)} In this case, the reaction is less selective. Nevertheless, the

products 547 obtained are very attractive candidates for medicinal chemistry.

The multicomponent reaction was used for the synthesis of tetrahydropyrimidines 548 .^{[193](#page-53-0)} The components were β -phenylamino- CF_3 -enone 549, primary amines 550, and formaldehyde 551. The reaction proceeds by heating in DMSO and leads to the tetrahydropyrimidines 548 in moderate-to-high yields.

 β -Alkoxy-CF₃-enones **552** were also used for the preparation of various CF_3 -pyrimidines 553 containing a 3-oxo-2,3-dihydropyrazole substituent.^{[194](#page-53-0)} These compounds are of particular interest because of their potential use as antiinflammatory non-steroidal agents. The starting 3-oxo-2,3 dihydropyrazole 554 derivative was prepared by the reaction of 3-acetylbutyrolactone 555 and aminoguanidine.

The reaction of β , β -dibromo-CF₃-ketone 13 with thioacetamide and thiourea was performed.[196](#page-53-0) The reaction was carried out in an alcoholic medium using a 100% excess of the thio compound. The products of the reaction—the corresponding 1,3-thiazine derivatives 560—were obtained in good yields. The reaction proceeds regioselectively, the nitrogen atom attacking the carbonyl group of 13.

Dihydrothiazine 561 can be prepared using the reaction of ketone 124 with 2-aminoethanethiol with subsequent oxidative cyclization of the adduct 562. The oxidative cyclization of 562 was performed by heating a DMSO solution in the presence of sodium carbonate.[178](#page-52-0)

4.1.3.4. Synthesis of 1,2-, 1,3-, and 1,4-thiazines. The reaction of β -alkoxy-CF₃-enones 556 with S,S-dimethylsulfoximine 557 was studied.[195](#page-53-0) Initially products 558 formed can be cyclized into the derivatives of 1,2-thiazine-1-oxide 559 in high yields.

4.1.3.5. Synthesis of 1,3-oxazines. The reaction of β -alkoxy-CF₃-enones **563** with ethyl carbamate leads to the formation of enamidoketones 565. Subsequent reduction to the aminoethanols 566 and cyclization to 567 was performed.[197](#page-53-0) One of the evaluated compounds 567 exhibited

significant in vitro activity against the tested microorganism strains.

4.1.4. Synthesis of seven-membered heterocycles.

4.1.4.1. Synthesis of 1,4-diazepines. 5-Trifluoromethyl-2,3-dihydro-1,4-diazepine 568 was prepared by the reaction of CF_3 -enone 124 with 1,2-propylenediamine 569. [44](#page-50-0) 1,2-Propylenediamine 569 reacted smoothly with 124 under mild reaction conditions to give 568. As expected, the reaction gave two isomeric products 568a and 568b in an almost 1:1 ratio. They can be separated cleanly by column chromatography.

4.1.5. Synthesis of condensed heterocycles.

4.1.5.1. Synthesis of quinoline derivatives. A method for the synthesis of CF_3 derivatives of dihydrobenzo[c]acridine 570 has been suggested.[198](#page-53-0) The approach is based on the application of β -methoxy CF₃-enone 571 obtained from tetralone. In the reaction of 571 with various substituted anilines, the formation of enaminoketones 572 is observed. The compounds 572 are cyclized into the target dihydrobenzo $[c]$ acridines 570 in high yields on treatment with polyphosphoric acid (PPA).

treated with polyphosphoric acid. Noteworthy the corresponding unsubstituted benzo[h]quinoline having $R_1=H$ cannot be prepared using this reaction.

An analogous approach was used in work^{[200](#page-53-0)} on the synthesis of 2-propyl-4-trifluoromethylquinoline 576. The enaminoketone 577 was prepared by the addition of aniline to the corresponding trifluoroacetylated propylacetylene.

Various enaminoketones 578 were used for the preparation of benzo[h]quinolines 579. The reaction with 1-aminonaphthalene was applied. The target heterocycles 579 were obtained in good yields using TFA as the cyclizing agent. 201

An effective method for preparation of 2-substituted 4 quinolinecarbaldehydes 580 was elaborated.^{[202](#page-53-0)} This method is based on the reaction of acetylenic ketone 245 with

The ketones 573 can be also used for the synthesis of benzo[h]quinolines 574 .^{[199](#page-53-0)} The intermediate enaminoketones 575 prepared from the enones 573 and 1-naphthylamine are 2-aminothiophenol 126. The reaction proceeds through the formation of a diacetal 581, which is hydrolyzed with formic acid. Unfortunately, the yield of 581 is only moderate.

4.1.5.2. Synthesis of other condensed heterocycles. The preparation of benzodiazepines 582 by a one-step synthesis using the reaction of ketones 583 with -phenylendiamine 118 in an ethanol/acetic acid mixture as the solvent has been reported.[203](#page-53-0) The yields of the target benzodiazepines 582 are sufficiently high.

A useful approach to the preparation of new CF_3 -containing 1,5-benzoxazepines 584 starting from o-aminophenols 585 and β -alkoxy ketone 224 was presented. An O–N exchange reaction gives the enaminoketones 586 in high yields. The reaction proceeds under mild conditions at room temperature. Further functionalization of 586 was achieved by heating with DMF-DMA. Thus, the corresponding dienaminoketones 587 were easily obtained. Cyclization of the resulting adducts 587 with aqueous sulfuric acid at 70 $\mathrm{^{\circ}C}$ occurred smoothly to give the fluorinated 1,5-benzoxazepines 584 in good yields. 204 204 204

The enone 124 reacts with 2-aminothiophenol 126 in toluene solution at room temperature forming the enaminoketone 588.^{[178](#page-52-0)} Heating of 588 in DMSO for 8 h leads to the product of oxidative cyclization—benzothiazine derivative 589—in high yield.

that destruction of the skeleton of the starting ketone 119 takes place.

The ketone 205 was applied for the synthesis of triazadibenzocrysenes 591. [205](#page-53-0) These polycondensed heterocycles containing various substituents were prepared in good yields in the reaction of ketone 205 with 2-perimidinylanilines 593.

The reaction of CF_3 -enone 42 with various amino derivatives of azoles 593—aminotriazoles, aminotetrazoles, and 2-aminobenzimidazoles was used for the preparation of dihydro-594 and tetrahydroazolopyrimidines 595.^{[206](#page-53-0)} In the case of aminotriazoles and aminotetrazoles, the reaction proceeds 100% stereoselectively to form compound 595 having a cis-orientation of the CF_3 and phenyl groups as a single diastereomer. 2-Aminobenzimidazole gives a diastereomeric mixture of 595 in a ratio of 3:1.

Treating enaminoketone 119 with methylamine or acetic acid leads to the formation of benzimidazole 590 containing a trifluoromethyl group in position $2⁴¹$ $2⁴¹$ $2⁴¹$ It should be noted The pyrimidine derivatives 596 can be prepared using the re-action of 71 with aminotriazoles and aminotetrazoles 593.^{[207](#page-53-0)} The intermediate tetrahydro derivatives 597 were obtained

as single diastereomers. The structure was determined by spectral data analysis. The formation of a single diastereomer was explained by high conformational energy of phenyl, ethoxycarbonyl, and CF_3 groups. As a result, the formation of more favorable diastereomer having an equatorial orientation of these substituents was observed.

Photo-induced cyclization of uracil-substituted ketones 167 having a sulfimino substituent was used for the preparation of pyrrolo[2,3-d]pyrimidine-2,4-diones 606 containing a CF_3 group.^{[61](#page-50-0)} The reaction proceeds at room temperature using a mercury lamp.

N

N $599(32%)$

Ph

O

The analogous reaction was investigated for the β -enamino-ketone 274.^{[208](#page-53-0)} This reaction leads directly to condensed heterocyclic compound 598 formation, bypassing the intermediate tetrahydro derivatives. The structure of the compounds 598 was determined by analysis of their NMR data. The reaction proceeds 100% regioselectively and the yields of the products 598 are good. The causes of the observed regioselectivity are not discussed.

The preparation of CF_3 derivatives of pyrido[2,3-d]pyrimidines 599 and 600, dihydro-derivative 601 and pyrazolo[3,4-b]pyridines 602 and 603 was studied.^{[209](#page-53-0)} The condensation of the corresponding 6-aminouracil 604 or 5-aminopyrazole 605 and CF₃-enones 42, 124, and 205 was used. The products were obtained in low yields, except for compounds 600.

The corresponding pyrido $[1,2-a]$ pyrimidine derivatives 607 are formed in good yields by heating the dienone 424 ob-tained from ketone 425 in toluene or acetic acid solution.^{[151](#page-52-0)}

It was shown that the reaction of ketone 124 with pyridinium (isoquinolinium) salts 608 in the presence of base leads to the indolizines 609. [210](#page-53-0) The prepared indolizines 609 were

N N

O

Me

obtained, due to the oxidation of the intermediate dihydroderivatives with atmospheric oxygen. In the case of the compound with $Y=COO$ Me, the partial transesterification of the product 609 is observed. Noteworthy is the fact that the reaction proceeds with low yields or does not proceed for the quinolinic salt.

N l
Me H O^ZN N CF₃

599 (32%) **601** (20%)

Ph

The reaction of ketones 222 (X=Br, I) with 2-aminopyridine was investigated.^{[211](#page-53-0)} The reaction leads in appropriate conditions to the imidazopyridine 610. In the case of compound 222 having $X=Cl$, the formation of a mixture of two

products is observed. One of the products is the imidazopyridine 610, and the second is the enaminoketone 611—the product of ethoxy-group substitution (the ratio of these compounds is \sim 1:1).

The synthesis of imidazopyridines 612 using the reaction of b-sulfonyl-substituted trifluoromethylketones 169 with several 2-aminopyridines 613 was described.^{[212](#page-53-0)} The reaction proceeds regio- and stereoselectively (the intermediate dihydro-derivatives 614 were isolated as a single diastereomer). This reaction is an exception in the commonly observed direction for the reaction of 169 with amines, because usually it leads to the products of sulfonyl-group substitution.^{[61](#page-50-0)} Noteworthy is the fact that the electrophilic attack is directed on C-3 of 169. Commonly, the C-4 carbon atom is the object of electrophilic attack for most of the α , β -unsaturated trifluoromethylketones. Such regiochemistry is observed probably due to the electronegative properties of the sulfonyl group.

The analogous regioselectivity is observed in the reaction of enones 169 with various 3-amino-1,2,4-triazoles and 5-ami-notetrazoles 619.^{[214](#page-53-0)} The reaction was carried out in conditions similar to those used in the pyrazolopyrimidine 616 synthesis. The 7-trifluoromethyl-substituted cycloadducts 620a dominate in most cases. A method for the synthesis of triazolopyrimidines 620b was elaborated. Such an inversion of selectivity was achieved by carrying out the reaction in acetonitrile.

The reaction of 2-amino-1,3,4-thiadiazoles 621 with 169, closely related to the reaction described above, was also investigated.[215](#page-53-0) Although the products 622 and 623 contain two asymmetric centers, the reaction proceeds with high stereoselectivity. The target compounds 622 and 623 were obtained in high yields. The structure of compound 622 having R=Ph and R_1 =H was established using X-ray analysis. The dihydropyrimidine cycle is positioned in an envelope conformation. The phenylsulfonyl group has an unusual axial orientation. This is probably due to an intramolecular

The synthesis of various heterocyclic systems using the reaction of enones 169 with several diazoles was investigated.^{[213](#page-53-0)} Reflux of 169 with 3-aminopyrazoles 615 leads to formation of the pyrazolopyrimidines 616a. In several cases, the isomeric pyrazolopyrimidines 616b were formed as the second product. Using aryl-substituted aminopyrazoles, the reaction proceed stereoselectively, forming 616a as the only isomer. In the reaction of ketones 169 with 2-amino-1H-benzimidazole 617, the formation of imidazopyridines 618 was observed.

hydrogen bond between the hydroxyl hydrogen atom and the phenylsulfonyl group oxygen atom.

The alkylation of pyridinethiones 458 and 459 with methyl iodide and ω -bromoacetophenones was studied.^{[163,164](#page-52-0)} The starting compounds were obtained from the ketones 224 and cyanoacetic acid thioamide. The corresponding methylthio and phenacylthio derivatives of nicotinonitrile 624 were obtained in good yields. The obtained compounds were also used for heterocyclization into the corresponding

benzoylthieno[2,3-b]pyridines 625 by treating 624 with potassium hydroxide in DMF solution.

a phenyl substituent. It was established that all substituents in this compound have an equatorial orientation.^{[177](#page-52-0)}

The same approach was applied for synthesis of thieno[2,3 b]pyridine 626 containing an N-phenylcarboxamide group instead of a benzoyl group.[216](#page-53-0) Chloroacetanilide was used in the second stage as the alkylating agent.

The trifluoromethylketone 154 can be applied for the preparation of electrophilic reagent 637, the vinylogous Vilsmeier-type reagent. The complex 637 can be used for different purposes. As an example, the reaction of

As a scaffold for the construction of condensed heterocyclic systems, several 2-aminothiazoles 627 were used.^{[217](#page-53-0)} The isomer 628a dominates over 628b among the products of this reaction. An attempt to use 2-amino-4-aryl-1,3-thiazoles failed because the reaction leads to predominate formation of the enaminoketones 629—the products of sulfonyl group nucleophilic substitution. In the reaction of benzothiazoles 630, the heterocycles 631 are formed as a single reaction product only in the case of compounds having the substituent in position 4, otherwise the formation of enaminoketone 632 is observed. In addition, the cyclization with 2-aminobenzothiazoles proceeds regio- and stereoselectively.

The reaction of ketones 633 with 2-mercaptobenzaldehyde 634 leads to the thiochromanes 635, which can be easily transformed into the 2H-thiochromenes 636 by heating the reaction mixture. The intermediate thiochromanes 635 were isolated only in the case of the CF_3 enone having

 $2,2'$ -bis-indolyl 638 with 637 leads to the formation of a pentacyclic compound 639. [218](#page-53-0) The reaction of N, N' -dipyrrolmethane 640 with 637 after alkaline treatment of the reaction mixture leads to aldehyde 641 formation.

The benzimidazolyl and benzoxazolyl CF_3 -ketones 642 and 643 were obtained in high yields in the reaction of o -phenylendiamine 118 or o -aminophenol 125 with β , β -dibromoketone $13.^{196}$ $13.^{196}$ $13.^{196}$

2-Trifluoromethylbenzimidazole 644 was prepared by the reaction of CF_3 -enone 124 with o-phenylenediamine 118. [44](#page-50-0) o-Phenylenediamine 118, the structure of which should favor 1,2-addition, reacted smoothly with 124. It did not, however, give the corresponding 2,3-diazepine derivative. Instead, the products were 2-trifluoromethylbenzimidazole 644 and benzimidazole 645.

4.2. Synthesis of carbocyclic systems

4.2.1. $[2+2]$ Addition reactions. It was found that bromotrifluoroacetylacetylene 257 reacts unexpectedly with alkenes $646a-h$ to give the $[2+2]$ cycloaddition adducts.^{[94](#page-51-0)} The reaction leads to the formation of trifluoroacetyl derivatives of cyclobutene 647a–h. These products were obtained in moderate-to-high yields. Noteworthy is the fact that [2+2] addition is the anomalous pathway, because it is forbidden according to Woodward–Hoffman orbital symmetry rules. This is explained by high electronegativity of the trifluoromethyl group. In addition, the formation of ene-reaction products is observed as by-products.

Analogously, chlorotrifluoroacetylacetylene 250 (X=Cl) [2+2] addition reactions with various vinyl ethers 648 have been carried out.²¹⁹ The reactions lead to the formation of mixtures of [2+2] cycloadducts 649 and ene-reaction products 650. A method for isolation of the cycloaddition product 649 was proposed based on a selective low-temperature bromination of 650.

The photo-induced cyclization of vinylacetylene 651 or tetra-methylethylene 652 with diketone 235 was studied.^{[86](#page-51-0)} The reaction was carried out until the conversion level of the starting enone became 40–45%. Mixtures of products 653, 654, 655, and 656 were obtained. The individual compounds can be isolated using column chromatography. The yields of the products are not given.

4.2.2. $[4+2]$ Addition reactions. Due to the presence of two EWG groups, ketones 169 can also be used as dienophiles in Diels–Alder reactions.[220](#page-53-0) The reaction of ketone 169 having $R=Ph$ with various dienes leads to the corresponding cyclohexenes 657 and bicycles 658 in high yields. The regio- and stereochemistries of compounds 657 and 658 were studied in detail. The elimination of sulfinic acids from the products of the Diels–Alder reactions 633 under treatment with DBU followed by oxidative aromatization of the 1,4-hexadienes 659 leads to the benzenes 660. Total resinification was observed for the reaction of

bicyclo[2.2.1]heptane derivatives 658. Nevertheless, in the case of bicyclo[2.2.2]octane derivatives 658, the corresponding dienes 661 were prepared in high yields.

The enone 663 obtained by oxidation of the corresponding sulfide 662 with the equivalent of 50% hydrogen peroxide in trifluoroacetic acid solution was used as the dienophile.^{[221](#page-53-0)} The compound 663 reacts readily with dienes 664 forming the intermediate adducts 665, which spontaneously eliminate the phenylsulfenic acid forming hexadienes 666 in high yields.

The reaction of ketone 663 with isoprene proceeds stereoselectively forming a mixture of isomers 667 and 668 in a 5:1 ratio. Activation of 663 by treatment with TFAA leads to the dienophile 669, which gives the opposite stereoselectivity the ratio of isomers $667/668$ becomes $1:1.5.^{221}$ $1:1.5.^{221}$ $1:1.5.^{221}$

The reaction of cyclopentadiene 39 with ketone 663 is not selective, forming four isomers of the adduct 670. The individual stereoisomers were not separated. The stereochemistry of the adducts 670 is discussed in detail.^{[221](#page-53-0)} In this case, elimination of a sulfinyl group did not take place. The results were explained by analyzing the theoretical calculations.

The influence of Lewis acids on the electrophilicity of carbonyl groups was investigated.[222](#page-53-0) It is well known that the coordination of carbonyl groups to Lewis acids exerts a dramatic effect on the rates and selectivities. While much research into Lewis acid-mediated stereoselective or regioselective reactions has been carried out, less attention has been paid to chemoselective reactions in the presence

of Lewis acids. It is evident that the more electrophilic aldehydes and ketones react with nucleophiles much faster than the less electrophilic analogs. It was found that the reverse is the case for Lewis acid-mediated reactions: the more electrophilic aldehydes and ketones react much slower than the less electrophilic analogs in the presence of Lewis acids, with a chemoselectivity that is not attainable under ordinary conditions. A similar observation was also made in a Diels– Alder reaction. The thermal reaction of a 1:1 mixture of ketones 671 and 42 with cyclopentadiene 39 at $40 °C$ afforded the [4+2] cycloadduct 673 derived from 42 in 90% yield, along with a small amount (8%) of 672 derived from 671. This is also an expected result, since 42, which has an electron-withdrawing CF_3 group, is a better dienophile than 671. The BF₃ · Et₂O-mediated reaction at -78 °C, however gave 673 as the sole product in 49% yield.

The Diels–Alder reaction for trimethylstannyl trifluoroacetylacetylene 256 and bromotrifluoroacetylacetylene 257 was investigated.^{[223](#page-53-0)} Using several model dienes 39, 674, and 682, the formation of the corresponding adducts 683 in high yields was demonstrated. In the case of the parent unsubstituted trifluoroacetylacetylene 257 (X=H), the Diels–Alder reaction gave the corresponding trifluoroacetyl derivatives of carbocyclic compounds 684–686 in high yields.^{[95](#page-51-0)}

Using the Diels–Alder reaction, a method for the preparation of substituted 2-trifluoroacetylbiphenyls 687 was elaborated.^{[224](#page-53-0)} This method is based on the Diels– Alder reaction of acetylenic CF_3 -ketones 245 and 688 with pyran derivatives 689 with subsequent acetone elimination.

4.2.3. Michael-based cyclizations. Trifluoromethylsubstituted arenes can be synthesized using the reaction of alkoxyenones 690 with various enamines 691 containing an EWG group in the β -position.^{[225](#page-53-0)} In one example, the reaction of ketones 690 with nitriles 691 allows the preparation of anilines 692. In the case of the compound having $R = COCF₃$, the cyclization and subsequent aromatization of dienones 693 proceed through the transition state 694 spontaneously. In the case of the compound with $R = CO₂Et$, the presence of triethylamine catalytic amounts is needed.

A similar reaction of enamines 704 with CF₃-enones 270 having aryl- and hetaryl substituents in the β -position was studied.^{[227](#page-53-0)} The corresponding bicyclic hydroxyketones 705 were obtained as a single diastereomer. The axial orientation of the hydroxy and aryl groups and the equatorial orientation of the CF_3 group was established using X-ray analysis.

The reactions of enone 124 with 1,3-dicarbonyl compounds 706 were studied. In addition to the previously

The formation of two regioisomers ('normal' 696 and the isomer 697) is observed in the case of the reaction of ketone 690 with enamines 695 containing an ethoxycarbonyl, acetyl or benzoyl group instead of a cyano group.[225](#page-53-0) The ratio of the products 696 and 697 varies over a wide range, but compound 696 predominates. The mechanism of formation of both products through the transition product 698 is reviewed in detail.

 $R = Ph$, 3-MeC₆H₄, 4-MeC₆H₄, 3-MeOC₆H₄, 2-thienyl, 2,5-(MeO)₂C₆

Work devoted to the construction of bicyclic systems containing trifluoromethyl substituents has been reported. The reaction of ketone 124 with 1-pyrrolidinocyclohexene 699 $(X=CH₂)$ leads to the formation of bicyclo[3.3.1]nonane-9-one derivative 700 ($X = CH₂$) as the single diastereomer. The analogous products were obtained in the case of the reactions of the ketone 124 with 1-pyrrolydino-4-methyl-cyclohexene and 1-pyrrolidinocycloheptene.^{[226](#page-53-0)}

Similarly, the ketone 124 reacts with enamines of non-cyclic ketones, e.g., 701. [226](#page-53-0) The cyclohexanone 702 obtained can be further dehydrated into 2,6-dimethyl-3-trifluoromethylphenol 703.

mentioned preparation of pyridine derivatives, the formation of primary substitution products 707 and the preparation of o-hydroxyacetophenone derivative 708 were demonstrated.^{[162](#page-52-0)}

4.3. Synthesis of alicyclic compounds

4.3.1. Reactions with C- and P-nucleophiles. The Baylis– Hillman reaction for acetylenic CF_3 -ketone 251 with acrolein, methyl vinyl ketone, methyl acrylate, and acrylonitrile

$$
E10
$$
\n
$$
E10
$$
\n
$$
E10
$$
\n
$$
E124
$$
\n
$$
M = \frac{1. Et_2O, 74 h, 20 °C}{2. AcoH (5% aq.)}
$$
\n
$$
F_3C
$$
\n
$$
H_2 = \frac{1. Et_2O, Et_3N, MsCl, 20 °C}{2. 2 M HCl aq., CH_2Cl_2}
$$
\n
$$
F_3C
$$
\n
$$
T03 (73%)
$$
\n
$$
T04
$$
\n
$$
T05
$$

was studied.^{[228](#page-53-0)} In the case of acrylonitrile, the target product has not been isolated. For the rest of the substrates, the corresponding products 709 were obtained in moderate yields.

The same reaction was studied for the CF_3 -enones 270 containing aryl substituents in the β -position.^{[229](#page-53-0)} Methyl acrylate, ethyl acrylate, methyl vinyl ketone, acrolein, phenyl vinyl sulfone, and acrylonitrile were studied as the activated alkenes. The reaction proceeds as a 1,2-addition of anion generated from the activated acrylonitrile to the carbonyl group of the CF_3 -ketone while using DABCO and acrylonitrile. The products of acrylonitrile 1,2-addition 710 were obtained in good yields. In the case of ketones containing EtO and Me2N groups only polymerization was observed and no target product was isolated. For the other activated alkenes, no reaction during a period of several months was observed.

The reaction of CF_3 -enones 270 containing various aryl substituents in the β -position with nitroalkanes was studied.^{[230](#page-53-0)} It was found that the reaction of α, β -unsaturated trifluoromethylketones 270 with ethyl nitroacetate proceeded in the presence of calcinated potassium fluoride (2 equiv) as the base in ethanol solution at room temperature to give the Michael adducts 711 in almost quantitative yields. Unexpected results were obtained while carrying out the reaction in aqueous ethanol in the presence of 1 equiv of KF. Instead of the Michael adduct, the product of the reaction was the corresponding $CF_3-\gamma$ -nitroketone 712 and spontaneous hydrolysis and decarboxylation reactions had taken place. The reaction was carried out with several α , β -unsaturated trifluoromethylketones and excellent results were obtained, the yields of the nitroketones 712 being almost quantitative. Therefore, the esters 711 and the unsubstituted ketones 712 have been prepared from the same reagents. Noteworthy is the fact that the corresponding CF_3 - γ -nitroketone 712 was obtained only in the case of CF_3 -enone containing a Ph substituent 270 (R=Ph) using the reaction with nitromethane.

The reaction of CF_3 -enones 270 with ethyl cyanoacetate has been studied.^{[231](#page-53-0)} Calcined potassium fluoride in i -propanol was used as the base. The reaction proceeds as a conjugated addition and the corresponding Michael adducts 713 were obtained in moderate yields. These compounds were formed as a mixture of diastereomers in approximately a 1:1 ratio. All attempts to perform the cyclization of the adducts 713 failed.

The reaction of enone 145 with benzylic Grignard reagents 714 in diethyl ether solution proceeds as a 1,2-addition,[232,233](#page-53-0) forming the corresponding alcohols 715, which can be transformed into the trifluoromethylnaphthalenes 716 in one step.[234](#page-53-0) Aryl and alkyl Grignard reagents give in a opposite manner the $1,4$ -adducts.^{[235](#page-53-0)} The analogous reaction of enone 333 with 714 proceeds non-selectively, and products similar to 715 were obtained in lower yields. Nevertheless, these products can also be transformed into the corresponding trifluoromethylnaphthalenes 716.^{[234](#page-53-0)}

The reaction of cyclic enone 145 with organozinc reagents was also studied.^{[236](#page-53-0)} The reaction proceeds as a 1,2-addition, forming, similar to 715, allylic alcohols 717 in good yields.

The reaction of ketones 718 with vinyl Grignard reagent 719 was carried out. Depending on the protective group, the reaction proceeds as a 1,4-conjugate addition or a mixture of 1,4-addition products 720 and 1,2-addition products 721 are formed.^{[75](#page-50-0)} In both cases, the yields of products are not high.

The reactions of 5-trifluoroacetyl-3,4-dihydro-2H-pyran 145 with various nucleophiles have been studied.^{[52](#page-50-0)} The ketone 145 reacts readily with Grignard reagents to give the ring-opened products. The corresponding alcohols 722 were obtained in good yields.

A 1,3-strain has been recognized as a major factor determining the stereochemical pathway of reactions in both cyclic and acyclic systems. The different course of addition of Grignard reagents to the ketone 386 where ring opening affords products, thus exposing a stereochemical control of A 1,3-strain, is reported.^{[237](#page-53-0)} Reaction of ketone 386 with arylmagnesium bromides gave diols 723 as the mixture of two diastereomers. The structure of the crystalline diastereomer of diol 723 ($Ar=4-MeC₆H₄$) was determined by single crystal X-ray diffraction. When the reaction is conducted with 3 equiv of 4-tolylmagnesium bromide at -35 °C the ketone 724 can be isolated in good yield. By effecting the reaction at 0° C, it is possible to isolate the aldehyde 725 in low yield.

Allylic alcohols are particularly useful intermediates for the synthesis of biologically active compounds. The trifluoromethylation reaction has been extended to the facile synthesis of trans-trifluoromethyl allylic alcohol 726 in excellent isolated yields by the CsF-catalyzed nucleophilic trifluoro-methylation of enone 42 with TMS-CF₃.^{[238](#page-53-0)} Initially, a study was carried out on the optimization of the reaction conditions. Using the optimized reaction conditions, trans enones were reacted with $TMS-CF_3$ in the presence of catalytic amounts of CsF to give the corresponding trifluoromethyl silyl ethers 727, hydrolysis of which with 6 N HCl at room temperature formed the trifluoromethyl allylic alcohols 726. In the case of ketone 42, the yields were almost quantitative.

The reaction of CF_3 -enone 124 with five different phospho-rous nucleophiles was studied.^{[239](#page-53-0)} In the case of triethyl phosphate, the oxaphospholene 728 is formed. In the case of $(EtO)₂P(O)H$, the oxaphospholene initially formed underwent ring opening and rearranged to a diethyl allyl phosphate 729 mixture. In the reaction of Bu_3P with 124, the initially formed anion attacked another molecule of 124 and finally gave the product 730. A stronger nucleophile, $(Et₂N)₃P$, gave 4-(diethylamino)-l,l,l-trifluoro-3-buten-2one 731 via nitrogen or phosphorus attack on the β -carbon atom of 124. The less reactive nucleophile Ph_3P did not, however, react with 124, even at higher temperatures and for longer reaction times.

4.3.2. Olefination of CF₃-enones. α , β -Unsaturated β -trifluoromethyl esters have found numerous applications in organic, materials, medicinal, and agricultural chemistry, owing to their unique physical, chemical, and biological properties. Huang et al. have investigated^{[240,241](#page-53-0)} olefination reactions of various carbonyl compounds using alkyl diazoacetates 371 and 732 in the presence of Fe and Co porphyrin complexes. These porphyrin derivatives are efficient and versatile catalysts for the selective olefination of a variety of carbonyl compounds including aldehydes and activated and inactivated ketones with diazo compounds. In the case of CF_3 -enone 42, the corresponding alkenes 733 having an E-configuration are formed in good yields.

The same ketone 42 was converted into the diene 734 using a Wittig reaction with complex phosphorus ylide 735.^{[242](#page-54-0)} The corresponding CF_3 -diene 734 was prepared in high yield.

The application of CF_3 -enone 124, its α -bromo derivative 222 and α -phenyl-substituted ketone 736 for the synthesis of various CF_3 -substituted derivatives of phthalic 737, 741,

4.3.3. Reduction reactions of CF₃-enones. For several CF₃enones 749 having a β -aryl substituent, the selective reduction of the carbonyl group has been performed.[244](#page-54-0) As the reducing reagent, sodium borohydride in the presence of cerium(III) chloride was used. The corresponding allylic alcohols 750 were obtained in high yields.

747, benzoic 738, 742, 745, 746, and picolinic acids 739 and 743 was described. 243 The method is based on the conversion of CF_3 -enones into the conjugated dienes 740, 744, and 748 using Wittig reactions and causing these dienes to react with the corresponding dienophiles.

 BF_4 MeO_2C S CF_3

735

MeO $_{2}$ C

42 735

Ph

The enantioselective reduction of enones 751 proceeds with the formation of allylic alcohols 752 as the only products.^{[245](#page-54-0)} The highest enantioselectivity was obtained in the case of DIP–chloride. This reagent allows the preparation of the alcohols 752, having an (S)-configuration. The configuration was established using X-ray analysis data for compound 752, where $R=4-BrC_6H_4$, as well as using ¹⁹F NMR

spectroscopy data of (S)-MPTA derivatives of alcohols 752. The application of a reducing reagent based on a mixture of catecholborane with the 20 mol % of substituted oxazaborolidine additive allows the synthesis of allylic alcohols 752 having an (R) -configuration, although the optical purity is lower than in the case of DIP-chloride application.

 $R= 4-BrC_6H_4$, $4-MeOC_6H_4$, $4-MeSC_6H_4$, 2 -naphthyl, $2-thienyl$

Using the example of acetylenic ketone 753, the possibility of chiral carbonyl group reduction was demonstrated.[246](#page-54-0) A chiral magnesium amide was used as the reducing agent. This reagent was generated in situ from di-i-propylmagnesium and the chiral amine 754. Despite the high ee ratio obtained in the case of non-fluorinated ketones, this system showed poor results in the reduction of the acetylenic CF_3 ketone—the high yield for the target propargyl alcohol 755 was 88%, although the ratio of enantiomers was only 1:2.

$$
Bu \xrightarrow{\text{Bu}} C_{F_3} \xrightarrow{i-Pr_2Mg/754}
$$
\n
$$
Bu \xrightarrow{\text{C}} C_{F_3} \xrightarrow{\text{C}} T55 (88\%)
$$
\n
$$
Ph \xrightarrow{\text{N}} N
$$
\n
$$
Ph \xrightarrow{\text{N}} N
$$
\n
$$
Pf_3 \xrightarrow{\text{C}} H T54
$$

4.3.4. CF_3 group elimination reactions. The work of Dekeyser and $David²⁴⁷$ $David²⁴⁷$ $David²⁴⁷$ is devoted to the synthesis of fungicidal carboxine analogs. The haloform reaction was carried out for cyclic α , β -dialkoxy-CF₃-enone 756 obtained by trifluoroacylation of 2-methyl-5,6-dihydrodioxine 757. The product of this reaction is 3-methyl-5,6-dihydrodioxine-2-carboxylic acid 758. The reactions described provide an example for the selective introduction of a carboxylic group using a trifluoroacylation/basic hydrolysis sequence.

5. Conclusions

Summarizing the facts given in this review, it might be concluded that elaboration of new of methods for the preparation of α , β -unsaturated trifluoromethylketones should be continued, although the trifluoroacylation of activated alkenes remains the most applicable method. The synthesis of some derivatives is hard and challenging work, but, nevertheless

these compounds possess very high synthetic potential as molecular building blocks containing the trifluoromethyl group. α, β -Unsaturated trifluoromethylketones are widely used, especially in the synthesis of heterocyclic compounds. The application of these very useful molecular building blocks is, however, not restricted by this area.

The peculiarities of α , β -unsaturated trifluoromethylketones are their high reactivity in the reactions with nucleophiles, as well as the high chemo-, regio-, and stereoselectivity in these reactions. The distinctive feature is the stability of the gem-hydroxy-trifluoromethyl fragments, sometimes very resistant to the action of dehydrating agents. It was demonstrated that gem-diol fragments, containing a trifluoromethyl group in the α -position, are also very stable.

References and notes

- 1. Nenaidenko, V. G.; Sanin, A. V.; Balenkova, E. S. Molecules 1997, 2, 186–232.
- 2. Nenaidenko, V. G.; Sanin, A. V.; Balenkova, E. S. Russ. Chem. Rev. 1999, 68, 437–458.
- 3. Flores, A. F. C.; Brondani, S.; Zanatta, N.; Rosa, A.; Martins, M. A. P. Tetrahedron Lett. 2002, 43, 8701–8705.
- 4. Bonacorso, H. G.; Martins, M. A. P.; Bittencourt, S. R. T.; Lourega, S. R. T.; Zanatta, N.; Flores, A. F. C. J. Fluorine Chem. 1999, 99, 177–182.
- 5. Vasil'tsov, A. M.; Shmidt, E. Yu.; Mikhaleva, A. I.; Zaitsev, A. B.; Tarasova, O. A.; Afonin, A. V.; Toryashinova, D.-S. D.; Il'icheva, L. N.; Trofimov, B. A. Russ. J. Org. Chem. 2001, 37, 334–338.
- 6. Zaitsev, A. B.; Shmidt, E. Yu.; Vasil'tsov, A. M.; Mikhaleva, A. I.; Morozova, L. V.; Ushakov, I. A.; Afonin, A. V.; Il'icheva, L. N. Russ. J. Org. Chem. 2003, 39, 1429–1435.
- 7. Levkovskaya, G. G.; Bozhenkov, G. V.; Larina, L. I.; Evstaf'eva, I. T.; Mirskova, A. N. Russ. J. Org. Chem. 2001, 37, 644–648.
- 8. Bozhenkov, G. V.; Levkovskaya, G. G.; Mirskova, A. N. Russ. J. Org. Chem. 2002, 38, 134–135.
- 9. Gorbunova, M. G.; Gerus, I. I.; Kukhar, V. P. Synthesis 2000, 738–742.
- 10. Gorlov, D. V.; Kurykin, M. A.; Petrova, O. E. Russ. Chem. Bull. 1999, 48, 1791–1792.
- 11. Kawase, M.; Hirabayashi, M.; Koiwai, H.; Yamamoto, K.; Miyamae, H. Chem. Commun. 1998, 641–642.
- 12. Vasileva, E.; Sapi, J.; Laronze, J.-Y.; Mirand, C.; Levy, J. Monatsh. Chem. 2002, 133, 151–156.
- 13. Sunose, M.; Anderson, K. M.; Orpen, A. G.; Gallagher, T.; Macdonald, S. J. F. Tetrahedron Lett. 1998, 39, 8885–8888.
- 14. Rashatasakhon, P.; Padwa, A. Org. Lett. 2003, 5, 189–191.
- 15. Padwa, A.; Rashatasakhon, P.; Rose, M. J. Org. Chem. 2003, 68, 5139–5146.
- 16. Cook, G.; Waddle, J. L. Tetrahedron Lett. 2003, 44, 6923– 6925.
- 17. Vronena, P. J. E.; Kovalb, N.; de Groota, A. ARKIVOC 2004, 24–50.
- 18. Simchen, G.; Schmidt, A. Synthesis 1997, 117–120.
- 19. Lamarre, C.; Stella, L. Synlett 1999, 725–726.
- 20. Liebeskind, L. S.; Srogl, J. J. Am. Chem. Soc. 2000, 122, 11260–11261.
- 21. Xiao, J.; Feng, Y.; Yuan, C. J. Chem. Soc., Perkin Trans. 1 2000, 4240–4241.
- 22. Fustero, S.; de la Torre, M. G.; Pina, B.; Fuentes, A. S. J. Org. Chem. 1999, 64, 5551–5556.
- 23. Sosnovskikh, V. Ya.; Usachev, B. I. Russ. Chem. Bull. 2004, 53, 383–392.
- 24. Zhang, L.; Brookhart, M.; White, P. S. Organometallics 2006, 25, 1868–1874.
- 25. Sosnovskikh, V. Ya.; Mel'nikov, B. I.; Kutsenko, V. A. Russ. J. Org. Chem. 1996, 32, 1393.
- 26. Reznikov, V. A.; Volodarsky, L. B. Russ. Chem. Bull. 1996, 45, 1699–1706.
- 27. Reznikov, V. A.; Roschupkina, G. I.; Mazhukin, D. G.; Petrov, P. A.; Popov, S. A.; Fokin, S. V.; Romanenko, G. V.; Rybalova, T. V.; Gatilov, Y. V.; Shvedenkov, Y. G.; Irtegova, I. G.; Shundrin, L. A.; Ovcharenko, V. I. Eur. J. Org. Chem. 2004, 749–765.
- 28. Gernert, D. L.; Ajamie, R. A.; Ardecky, R. A.; Bell, M. G.; Leibowitz, M. D.; Mais, D. A.; Mapes, C. M.; Michellys, P. Y.; Rungta, D.; Reifel-Miller, A.; Tyhonas, J. S.; Yumibe, N.; Grese, T. A. Bioorg. Med. Chem. Lett. 2003, 13, 3191– 3195.
- 29. Pryadeina, M. V.; Kuzueva, O. G.; Burgart, Ya. V.; Saloutin, V. I.; Lyssenko, K. A.; Antipi, M. Yu J. Fluorine Chem. 2002, 117, 1–7.
- 30. Palanki, M. S. S.; Gayo-Fung, L. G.; Shevlin, G. I.; Erdman, P.; Sato, M.; Goldman, M.; Ransone, L.; Sponner, C. Bioorg. Med. Chem. Lett. 2002, 12, 2573–2577.
- 31. Yilmaz, M.; Pekel, A. T. J. Fluorine Chem. 2005, 126, 401– 406.
- 32. Lee, H.-S.; Kim, K. Tetrahedron Lett. 1998, 39, 5781–5784.
- 33. Anshworth, I.; Hopes, P.; Levin, D.; Patel, I.; Salloo, R. Tetrahedron Lett. 2002, 43, 4931–4933.
- 34. Palacios, F.; Pascual, S.; Oyarzabal, J.; de Retana, A. M. O. Org. Lett. 2002, 4, 769–772.
- 35. Palacios, F.; Oyarzabal, J.; de Retana, A. M. O.; Pascual, S.; Oyarzabal, J. J. Org. Chem. 2004, 69, 8767–8774.
- 36. Yoshimatsu, M.; Sugimoto, T.; Okada, N.; Kinoshita, S. J. Org. Chem. 1999, 64, 5162–5165.
- 37. Matsubara, Y.; Yoshimatsu, M. J. Org. Chem. 2000, 65, 4456– 4459.
- 38. Yoshimatsu, M.; Timura, Y. J. Org. Chem. 2002, 67, 5678– 5682.
- 39. Jiang, B. Chem. Commun. 1996, 861–862.
- 40. Peng, W.; Zhu, S. Tetrahedron 2003, 59, 4641–4649.
- 41. Pashkevich, K. I.; Khomutov, O. G.; Sevenard, D. V. Russ. Chem. Bull. 1999, 48, 557–560.
- 42. Venkat Reddy, G.; Rama Rao, V. V. V. N. S.; Maitraie, D.; Ravikanth, S.; Yadla, R.; Reddy, S. N.; Narsaiah, B.; Shanthan Rao, P. J. Fluorine Chem. 2003, 124, 203–209.
- 43. Yachevskii, D. S.; Chizhov, D. L.; Kodess, M. I.; Pashkevich, K. I. Monatsh. Chem. 2004, 135, 23–30.
- 44. Zhu, S.; Chu, Q.; Wang, Y. Heteroat. Chem. 2000, 11, 27– 30.
- 45. Sosnovskikh, V. Ya. Russ. Chem. Bull. 1997, 46, 2145–2146.
- 46. Sosnovskikh, V. Ya.; Mel'nikov, M. Yu.; Zaitsev, A. V.; Bogdanov, E. A. Russ. Chem. Bull. 1998, 47, 1170–1174.
- 47. Cebulska, Z.; Laurent, A. J.; Laurent, E. G. J. Fluorine Chem. 1996, 76, 177–180.
- 48. Pashkevich, K. I.; Khomutov, O. G.; Sevenard, D. V. Russ. J. Org. Chem. 1998, 34, 1727–1730.
- 49. Bartoli, G.; Bosco, M.; Locatelli, M.; Marcantoni, E.; Melchiorre, P.; Sambri, L. Synlett 2004, 239–242.
- 50. Ohkura, H.; Berbasov, D. O.; Soloshonok, V. A. Tetrahedron 2003, 59, 1647–1656.
- 51. Zanatta, N.; Squizani, A. M. C.; Fantinel, L.; Nachtigall, F. M.; Bonacorso, H. G.; Martins, M. A. P. Synthesis 2002, 2409–2415.
- 52. Zhu, S.; Xu, G.; Qin, C.; Chu, Q.; Xu, Y. Monatsh. Chem. 1999, 130, 671–680.
- 53. Arcadi, A.; Bianchi, G.; Di Giuseppe, S.; Marinelli, F. Green Chem. 2003, 5, 64–67.
- 54. Singh, R. P.; Twamley, B.; Shreeve, J. M. J. Org. Chem. 2002, 67, 1918–1924.
- 55. Andrew, R. J.; Mellor, J. M. Tetrahedron 2000, 56, 7261– 7266.
- 56. Sanin, A. V.; Nenaidenko, V. G.; Denisenko, D. I.; Smolko, K. I.; Balenkova, E. S. Russ. J. Org. Chem. 1999, 35, 209–211.
- 57. Koldobskii, A. B.; Tsvetkov, N. P.; Godovikov, I. A.; Kalinin, V. N. J. Fluorine Chem., in press.
- 58. Koldobskii, A. B.; Tsvetkov, N. P.; Godovikov, I. A.; Kalinin, V. N. J. Fluorine Chem., in press.
- 59. Koldobskii, A. B.; Tsvetkov, N. P.; Peregudov, A. S.; Kalinin, V. N. J. Fluorine Chem., in press.
- 60. Martins, M. A. P.; Emmerich, D. J.; Pereira, C. M. P.; Cunico, W.; Rossato, M.; Zanatta, N.; Bonacorso, H. Tetrahedron Lett. 2004, 45, 4935–4938.
- 61. Matsumoto, N.; Takahashi, M. Tetrahedron Lett. 2005, 46, 5551–5554.
- 62. Krasovsky, A. L.; Nenajdenko, V. G.; Balenkova, E. S. Russ. Chem. Bull. 2001, 50, 1395–1400.
- 63. Krasovsky, A. L.; Nenajdenko, V. G.; Balenkova, E. S. Russ. Chem. Bull. 2002, 51, 2080–2085.
- 64. Nenajdenko, V. G.; Krasovsky, A. L.; Lebedev, M. V.; Balenkova, E. S. Synlett 1997, 1349–1350.
- 65. Krasovsky, A. L.; Pisarev, S. A.; Nenajdenko, V. G.; Balenkova, E. S. Russ. Chem. Bull. 2003, 52, 1791–1796.
- 66. Krasovsky, A. L.; Druzhinin, S. V.; Nenajdenko, V. G.; Balenkova, E. S. Tetrahedron Lett. 2004, 45, 1129–1132.
- 67. Funabiki, K.; Ohtsuki, T.; Ishira, T.; Yamanaka, H. J. Chem. Soc., Perkin Trans. 1 1998, 2413–2423.
- 68. Yang, X.-J.; Liu, J.-T.; Zhao, F.-L. J. Fluorine Chem. 2004, 125, 415–419.
- 69. Martins, M. A. P.; Bastos, G. P.; Sinhorin, A. P.; Zimmermann, N. E. K.; Bonacorso, H. G.; Zanatta, N. Synthesis 2002, 15, 2220–2224.
- 70. Stefani, H. A.; Pereire, C. M. P.; Doerr, F. A.; Cella, R. ARKIVOC 2005, 6, 19–24.
- 71. Gerus, I. I.; Kruchok, I. S.; Kukhar, V. P. Tetrahedron Lett. 1999, 40, 5923–5926.
- 72. Kruchok, I. S.; Gerus, I. I.; Kukhar, V. P. Synthesis 2002, 71–74.
- 73. Kruchok, I. S.; Gerus, I. I.; Kukhar, V. P. Tetrahedron 2000, 56, 6533–6539.
- 74. Singh, R. P.; Cao, G.; Kirchmeier, R. L.; Shreeve, J. M. J. Org. Chem. 1999, 64, 2873–2876.
- 75. Rodeschini, V.; Van de Weghe, P.; Salomon, E.; Tarnus, C.; Eustache, J. J. Org. Chem. 2005, 70, 2409–2412.
- 76. Surya Prakash, G. K.; Mandal, M.; Schweizer, S.; Petasis, N. A.; Olah, G. A. Org. Lett. 2000, 2, 3173–3176.
- 77. Baxendalea, I. R.; Ley, S. V.; Lumerasb, W.; Nesic, M. Comb. Chem. High Throughput Screening 2002, 5, 197–199.
- 78. Ishizaki, M.; Suzuki, D.; Hoshino, O. J. Fluorine Chem. 2001, 81–90.
- 79. Gerus, I. I.; Kacharova, L. M.; Vdovenko, S. I. Synthesis 2001, 431–436.
- 80. Kacharova, L. M.; Kacharov, A. D.; Gerus, I. I. J. Fluorine Chem. 2001, 111, 29–31.
- 81. Reznikov, V. A.; Skuridin, N. G.; Khromovskikh, E. L.; Khramtsov, V. V. Russ. Chem. Bull. 2003, 52, 2052–2056.
- 82. Rulev, A. Yu.; Fedorov, S. V.; Nenajdenko, V. G.; Balenkova, E. S.; Voronkov, M. G. Russ. Chem. Bull. 2003, 52, 2287– 2289.
- 83. Fedorov, S. V.; Rulev, A. Yu.; Chipanina, N. N.; Shulunova, A. M.; Nenajdenko, V. G.; Balenkova, E. S.; Tyurin, D. A.; Turchaninov, V. K. Russ. Chem. Bull. 2005, 54, 103–107.
- 84. Lyutenko, N. V.; Gerus, I. I.; Kacharov, A. D.; Kukhar, V. P. Tetrahedron 2003, 59, 1731–1738.
- 85. Gerus, I. I.; Lyutenko, N. V.; Kacharov, A. D.; Kukhar, V. P. Tetrahedron Lett. 2000, 41, 10141–10145.
- 86. Ferrer, L. O.; Margaretha, P. Chem. Commun. 2001, 481–482.
- 87. Hegde, V. B.; Renga, J. M.; Owen, J. M. Tetrahedron Lett. 2001, 1847–1849.
- 88. Jones, B. G.; Branch, S. K.; Thompson, A. S.; Threadgill, M. D. J. Chem. Soc., Perkin Trans. 1 1996, 2685–2691.
- 89. Frey, R. R.; Wada, C. K.; Garland, R. B.; Curtin, M. L.; Michaelides, M. R.; Li, J.; Pease, L. J.; Glaser, K. B.; Marcotte, P. A.; Bouska, J. J.; Murphy, S. S.; Davidsen, S. K. Bioorg. Med. Chem. Lett. 2002, 12, 3443–3447.
- 90. Obrecht, D.; Gerber, F.; Sprenger, D.; Masquelin, T. Helv. Chim. Acta 1997, 80, 531–537.
- 91. Lantzsch, R.; Himmler, T.; Marhold, A. DE Patent 19,621,687 A1, 1997.
- 92. Blay, G.; Fernandez, I.; Marco-Aleixandre, A.; Monje, B.; Pedro, J. R.; Ruiz, R. Tetrahedron 2002, 58, 8565–8571.
- 93. Koldobskii, A. B.; Solodova, E. V.; Kalinin, V. N. Dokl. Chem. 1999, 366, 1–3, 110–112.
- 94. Tsvetkov, N. P.; Koldobskii, A. B.; Kalinin, V. N. Dokl. Chem. 2005, 404, 201–204.
- 95. Tsvetkov, N. P.; Koldobskii, A. B.; Korzinkova, A. S.; Peregudov, A. S.; Kalinin, V. N. Dokl. Chem. 2006, 408, 481–482.
- 96. Tsvetkov, N. P.; Koldobskii, A. B., unpublished results.
- 97. Coe, P. L.; Owen, I. R.; Till, S. J. J. Chem. Soc., Perkin Trans. 1 2000, 1529–1535.
- 98. Martins, M. A. P.; Pereira, C. M. P.; Zimmermann, N. E. K.; Cunico, W.; Moura, S.; Beck, P.; Zanatta, N.; Bonacorso, H. G. J. Fluorine Chem. 2003, 123, 261–265.
- 99. Zanatta, N.; Rosa, L. S.; Cortelini, M. F. M.; Beux, S.; Santos, A. P. D.; Bonacorso, H. G.; Martins, M. A. P. Synthesis 2002, 2404–2408.
- 100. Zanatta, N.; Rosa, L. S.; Loro, E.; Bonacorso, H. G.; Martins, M. A. P. J. Fluorine Chem. 2001, 107, 149–154.
- 101. Nenaidenko, V. G.; Druzhinin, S. V.; Balenkova, E. S. Russ. Chem. Bull. 2003, 52, 11 2467–2472.
- 102. Lee, H.-S.; Kim, K. Tetrahedron Lett. 1998, 39, 6895–6898.
- 103. Cebulska, Z.; Laurent, A. J.; Laurent, E. G. Bull. Soc. Chim. Fr. 1996, 133, 209–212.
- 104. Krasnykh, O. P.; Karpenko, N. S.; Filyakova, V. I.; Charushin, V. N. Russ. Chem. Bull. 2004, 53, 1355–1356.
- 105. Zanatta, N.; Schneider, J. M. F.; Schneider, P. H.; Wouters, A. D.; Bonacorso, H. G.; Martins, M. A. P.; Wessjohann, L. A. J. Org. Chem. 2006, 71, 6996–6998.
- 106. Andrew, R. J.; Mellor, J. M. Tetrahedron 2000, 56, 7267– 7272.
- 107. Rudenko, A. P.; Aristov, S. A.; Vasilyev, A. V. Russ. J. Org. Chem. 2004, 40, 1221.
- 108. Aristov, S. A.; Vasilyev, A. V.; Rudenko, A. P. Russ. J. Org. Chem. 2006, 42, 66–72.
- 109. Touzot, A.; Soufyane, M.; Berber, H.; Toupet, L.; Mirand, C. J. Fluorine Chem. 2004, 125, 1299–1304.
- 110. Lui, N.; Brackemeyer, T.; Muller, P.; Schneder, M. WO Patent 03,051,820, 2003.
- 111. Pashkevich, K. I.; Filyakova, V. I.; Kuznetsova, O. A. Russ. Chem. Bull. 1996, 45, 2868–2870.
- 112. Pavlik, J. W.; Ayudhya, T. I. N.; Tantaynon, S. J. Heterocycl. Chem. 2002, 39, 1025–1027.
- 113. Schlosser, M.; Volle, J.-N.; Leroux, F.; Schenk, K. Eur. J. Org. Chem. 2002, 2913–2920.
- 114. Pavlik, J. W.; Ayudhya, T. I. N.; Tantaynon, S. J. Heterocycl. Chem. 2003, 40, 1087–1089.
- 115. Martins, M. A. P.; Cunico, W.; Brondani, S.; Peres, R. L.; Zimmerman, N.; Rosa, F. A.; Fiss, G. F.; Zanatta, N.; Bonacorso, H. G. Synthesis 2006, 1485–1493.
- 116. Singh, S. P.; Kumar, D. J. Chem. Res., Synop. 1997, 142–143.
- 117. Martins, M. A. P.; Blanco, R. F.; Pereira, C. M. P.; Beck, P.; Brondani, S.; Cunico, W.; Zimmermann, N. E. K.; Bonacorso, H. G.; Zanatta, N. J. Fluorine Chem. 2002, 118, 69–72.
- 118. Nenajdenko, V. G.; Reznichenko, A. L.; Balenkova, E. S. Russ. Chem. Bull. 2006, 1, 172.
- 119. Levskovskaya, G. G.; Bozhenkov, G. V.; Larina, L. I.; Mirskova, A. N. Russ. J. Org. Chem. 2002, 38, 1501–1506.
- 120. Bozhenkov, G. V.; Levskovskaya, G. G.; Mirskova, A. N.; Dolgushin, G. V.; Larina, L. I.; Ushakov, P. E. Russ. J. Org. Chem. 2003, 39, 1069–1075.
- 121. Martins, M. A. P.; Bastos, G. P.; Sinhorin, A. P.; Zimmermann, N. E. K.; Rosa, A.; Brondani, S.; Emmerich, D.; Bonacorso, H. G.; Zanatta, N. J. Fluorine Chem. 2003, 123, 249–253.
- 122. Flores, A. F. C.; Brondani, S.; Pizutti, L.; Martins, M. A. P.; Zanatta, N.; Bonacorso, H. G. Synthesis 2005, 2744–2750.
- 123. Song, L.-P.; Chu, Q.-L.; Zhu, S.-Z. J. Fluorine Chem. 2001, 107, 107–112.
- 124. Kawase, M.; Hyrabayashi, M.; Saito, S.; Yamamoto, K. Tetrahedron Lett. 1999, 40, 2541–2544.
- 125. Mellor, J. M.; Schofield, S. R.; Korn, S. R. Tetrahedron 1997, 53, 17163–17170.
- 126. Bonacorso, H. G.; Cechinel, C. A.; Oliveira, M. R.; Costa, M. B.; Martins, M. A. P.; Zanatta, N.; Flores, A. F. C. J. Heterocycl. Chem. 2005, 42, 1055–1061.
- 127. Cunico, W.; Cleber, A.; Cechinel, A.; Bonacorso, H. G.; Martins, M. A. P.; Zanatta, N.; de Souza, M. V. N.; Fraitas, I. O.; Soares, R. P. P.; Kretti, A. U. Bioorg. Med. Chem. Lett. 2006, 16, 649–653.
- 128. Bonacorso, H. G.; Wentz, A. P.; Zanatta, N.; Martins, M. A. P. Synthesis 2001, 1505–1508.
- 129. Zanatta, N.; Flores, D. C.; Madruga, C. C.; Faoro, D.; Flores, A. F. C.; Bonacorso, H. G.; Martins, M. A. P. Synthesis 2003, 894–898.
- 130. Pryadeina, M. V.; Burgart, Ya. V.; Kodess, M. I.; Saloutin, V. I.; Chupakhin, O. N. Russ. Chem. Bull. 2004, 53, 1261– 1266.
- 131. Bonacorso, H. G.; Wastowski, A. D.; Zanatta, N.; Martins, M. A. P.; Naue, J. A. J. Fluorine Chem. 1998, 92, 23–26.
- 132. Martins, M. A. P.; Sinhorin, A. P.; da Rosa, A.; Flores, A. F. C.; Wastowski, A. D.; Pereira, C. M. P.; Flores, D. C.; Beck, P.; Freitag, R. A.; Brondani, S.; Cunico, W.; Bonacorso, H. G.; Zanatta, N. Synthesis 2002, 2353– 2358.
- 133. Bonacorso, H. G.; Lewandrowski, H.; Drekener, R. L.; Costa, M. B.; Pereira, C. M. P.; Wastowski, A. D.; Peppe, C.; Martins, M. A. P.; Zanatta, N. J. Fluorine Chem. 2003, 122, 159–163.
- 134. Bonacorso, H. G.; Lang, E. S.; Lewandowski, H.; Martins, M. A. P.; Peppe, C.; Zanatta, N. Inorg. Chem. Commun. 2003, 6, 646–649.
- 135. Reddy, M.; Bell, S. WO Patent 03,024,958 A2, 2003.
- 136. Reddy, M.; Bell, S. WO Patent 03,024,400 A2, 2003.
- 137. Nenajdenko, V. G.; Statsuk, A. V.; Balenkova, E. S. Chem. Heterocycl. Compd. 2003, 5, 598–603.
- 138. Trofimov, B. A.; Schmidt, E. Yu.; Mikhaleva, A. I.; Vasil'tsov, A. M.; Larina, L. I.; Klyba, L. V. Mendeleev Commun. 1999, 6, 238–239.
- 139. Martins, M. A. P.; Siqueira, G. M.; Bastos, G. P.; Bonacorso, H. G.; Zanatta, N. J. Heterocycl. Chem. 1996, 33, 1619– 1622.
- 140. Okada, E.; Okumura, H.; Nishida, Y.; Kitahora, T. Heterocycles 1999, 50, 377–384.
- 141. Martins, M. A. P.; Flores, A. F. C.; Bastos, G. P.; Zanatta, N.; Bonacorso, H. G. J. Heterocycl. Chem. 1999, 36, 837–840.
- 142. Zhu, S.; Jin, G.; Peng, W.; Huang, Q. Tetrahedron 2003, 59, 2899–2905.
- 143. Majo, V. J.; Prabhakaran, J.; Simpson, N. R.; Van Heertum, R. L.; Mann, J. J.; Dileep Kumar, J. S. Bioorg. Med. Chem. Lett. 2005, 15, 4268–4271.
- 144. Soloshonok, V. A.; Kacharov, A. D.; Avilov, D. V. Tetrahedron Lett. 1996, 43, 7845–7848.
- 145. Soloshonok, V. A.; Kacharov, A. D.; Avilov, D. V.; Ishikawa, K.; Nagashima, N.; Hayashi, T. J. Org. Chem. 1997, 62, 3470– 3479.
- 146. Peng, W.-M.; Zhu, S.-Z. J. Fluorine Chem. 2002, 116, 81– 86.
- 147. Cooke, J. W. B.; Coleman, M. J.; Caine, D. M.; Jenkins, K. P. Tetrahedron Lett. 1998, 39, 7965–7966.
- 148. Cocco, M. T.; Congiu, C.; Onnis, V.; Morelli, M.; Felipo, V.; Cauli, O. Bioorg. Med. Chem. Lett. 2004, 12, 4169– 4177.
- 149. Cocco, M. T.; Congiu, C.; Onnis, V. J. Heterocycl. Chem. 1996, 33, 1771–1773.
- 150. Cocco, M. T.; Congiu, C.; Onnis, V. Tetrahedron Lett. 1999, 40, 4407–4410.
- 151. Cocco, M. T.; Congiu, C.; Onnis, V. Chem. Pharm. Bull. 2001, 49, 703–706.
- 152. Cocco, M. T.; Congiu, C.; Lilliu, V.; Onnis, V. Bioorg. Med. Chem. Lett. 2004, 14, 5787–5791.
- 153. Cocco, M. T.; Congiu, C.; Lilliu, V.; Onnis, V. J. Med. Chem. 2005, 48, 8245–8252.
- 154. Carroll, W. A.; Altenbach, R. J.; Bai, H.; Brioni, J. D.; Brune, M. E.; Buckner, S. A.; Cassidy, C.; Chen, Y.; Coghlan, M. J.; Daza, A. V.; Drizin, I.; Fey, T. A.; Fitzgerald, M.; Gopalakrishnan, M.; Gregg, R. J.; Henry, R. F.; Holladay, M. W.; King, L. L.; Kort, M. E.; Kym, P. R.; Milicic, I.; Tang, R.; Turner, S. C.; Whiteaker, K. L.; Yi, L.; Zhang, H.; Sullivan, J. P. J. Med. Chem. 2004, 47, 3163–3179.
- 155. Konakahara, T.; Hojahmat, M.; Tamura, S. J. Chem. Soc., Perkin Trans. 1 1999, 2803–2806.
- 156. Konakahara, T.; Sugama, N.; Yamada, A.; Kakehi, A.; Sakai, N. Heterocycles 2001, 55, 313–322.
- 157. Hojahmat, M.; Konakahara, T.; Tamura, S. Heterocycles 2000, 53, 629–636.
- 158. Okada, E.; Kinomura, T.; Higashiyama, Y.; Takeuchi, H.; Hojo, M. Heterocycles 1997, 44, 129–132.
- 159. Lahm, G. P.; Selby, T. P.; Freudenberger, J. H.; Stevenson, T. M.; Myers, B. J.; Seburyamo, G.; Smith, B. K.; Flexner, L.; Clark, C. E.; Cordova, D. Bioorg. Med. Chem. Lett. 2005, 15, 4898–4906.
- 160. Funabiki, K.; Isomura, A.; Yamaguchi, Y.; Hashimoto, W.; Matsumaga, K.; Shibata, K.; Matsui, M. J. Chem. Soc., Perkin Trans. 1 2001, 2578–2592.
- 161. Isaacs, R. C. A.; Cutrona, K. J.; Newton, C. L.; Sanderson, P. E.; Solinski, M. G.; Baskin, E. P.; Chen, I.-Wu.; Cooper, C. M.; Cook, J. J.; Cardell, S. J.; Lews, S. D.; Lucas, R. J., Jr.; Lyle, E. A.; Lynch, J. J., Jr.; Naylor-Olsen, A. M.; Stranieri, M. T.; Vastag, K.; Vacca, J. P. Bioorg. Med. Chem. Lett. 1998, 8, 1719–1724.
- 162. Zanatta, N.; Barichello, R.; Bonacorso, H. G.; Martins, M. A. P. Synthesis 1999, 765–768.
- 163. Kislyi, V. P.; Nikishin, K. G.; Kruglova, E. Ya.; Shestopalov, A. M.; Semenov, V. V. Tetrahedron 1996, 33, 10841–10848.
- 164. Yakunin, Ya. Yu.; Dyachenko, V. D.; Litvinov, V. P. Chem. Heterocycl. Compd. 2000, 36, 1431–1436.
- 165. Nenajdenko, V. G.; Druzhinin, S. V.; Balenkova, E. S. J. Fluorine Chem. 2006, 127, 856–873.
- 166. Koyanagi, T.; Kusatsu-shi, S.; Yoneda, T.; Kanamori, F.; Kanbayashi, S.; Tnimura, T.; Horiuchi, N. EP Patent 0,744,400 A2.
- 167. Bayer Cropscience GMBP. WO Patent 03,022,818 A1, 2003.
- 168. Aventis Cropscience GMBP. WO Patent 0,248,111 A2, 2002.
- 169. Hamilton, C. T.; Gullo, M. F.; Gonzalez, M. A.; Roth, G. A.; Gorman, D. B. U.S. Patent Appl. 20,050,288,511 A1, 2005.
- 170. Xiong, W.-N.; Yang, C.-G.; Jiang, B. Bioorg. Med. Chem. Lett. 2001, 9, 1773–1780.
- 171. Jiang, B.; Xiong, W.; Zhang, X.; Zhang, F. Org. Process Res. Dev. 2001, 5, 531–534.
- 172. Amii, H.; Kobayashi, T.; Terasawa, H.; Uneyama, K. Org. Lett. 2001, 3, 3103-3105.
- 173. Krasovsky, A. L.; Nenajdenko, V. G.; Balenkova, E. S. Russ. Chem. Bull. 2002, 51, 609–612.
- 174. Hayman, C. M.; Larsen, D. S.; Brooker, S. Aust. J. Chem. 1998, 51, 545–553.
- 175. Hayman, C. M.; Hanton, L. R.; Larsen, D. S.; Guthrie, J. M. Aust. J. Chem. 1999, 52, 921–927.
- 176. Nenajdenko, V. G.; Druzhinin, S. V.; Balenkova, E. S. Mendeleev Commun. 2006, 16, 180–182.
- 177. Nenajdenko, V. G.; Sanin, A. V.; Churakov, A. V.; Howard, J. A. K.; Balenkova, E. S. Chem. Heterocycl. Compd. 1999, 35, 549–556.
- 178. Chu, Q.; Song, L.; Jin, G.; Zhu, S. J. Fluorine Chem. 2001, 108, 51–56.
- 179. Gerus, I. I.; Tolmacheva, N. A.; Vdovenko, S. I.; Frohlich, R.; Haufe, G. Synthesis 2005, 1269–1278.
- 180. Bonacorso, H. G.; Lopes, I. S.; Wastowski, A. D.; Zanatta, N.; Martins, M. A. P. J. Fluorine Chem. 2003, 120, 29–32.
- 181. Zanatta, N.; Fagundes, M. B.; Ellensohn, R.; Marcues, M.; Bonacorso, H. G.; Martins, M. A. P. J. Heterocycl. Chem. 1998, 35, 451–454.
- 182. Funabiki, K.; Nakamura, H.; Matsui, M.; Shibata, K. Synlett 1999, 756–758.
- 183. Zanatta, N.; Madruga, C. D. C.; Marisco, P. D. C.; Florres, D. C.; Bonacorso, H. G.; Martins, M. A. P. J. Heterocycl. Chem. 2000, 37, 1213–1218.
- 184. Berber, H.; Soufyane, M.; Santillana-Hayat, M.; Miranda, C. Tetrahedron Lett. 2002, 43, 9233–9235.
- 185. Cechin, S. R.; Schetinger, M. R. C.; Zanatta, N.; Madruga, C. C.; Pacholski, I. L.; Flores, D. C.; Bonacorso, H. G.; Martins, M. A. P.; Morsch, V. M. Chem. Res. Toxicol. 2003, 16, 1433–1439.
- 186. Palanki, M. S. S.; Erdman, P. E.; Gayo-Fung, L. M.; Shevlin, G. I.; Sullivan, R. W.; Suto, M. J.; Goldman, M. E.; Ransone,

L. J.; Brydon, B. L.; Manning, A. M. J. Med. Chem. 2000, 43, 3995–4004.

- 187. Ondi, L.; Lefebvre, O.; Schlosser, M. Eur. J. Org. Chem. 2004, 3714–3718.
- 188. Zanatta, N.; Lopes, E. C. S.; Fantinel, L.; Bonacorso, H. G.; Martins, M. A. P. J. Heterocycl. Chem. 2002, 39, 943–947.
- 189. Dorigo, P.; Fraccarollo, D.; Santostasi, G.; Marango, I.; Floreani, M.; Borea, P. A.; Mosti, L.; Sansebastiano, L.; Fossa, P.; Orsini, F.; Benetollo, F.; Bombrieri, G. J. Med. Chem. 1996, 39, 3671–3683.
- 190. Soufyane, M.; van den Brock, S.; Khamliche, L.; Mirand, C. Heterocycles 1999, 10, 2445–2451.
- 191. Okada, E.; Kinomura, T.; Takeuchi, H.; Hojo, M. Heterocycles 1997, 44, 349–356.
- 192. Bonacorso, H. G.; Costa, M. B.; Lopes, I. S.; Oliveira, M. R.; Drekener, R. L.; Martins, M. A. P.; Zanatta, N.; Flores, A. F. C. Synth. Commun. 2005, 35, 3055–3064.
- 193. Zhao, F.-L.; Liu, J.-T. J. Fluorine Chem. 2004, 125, 1491– 1496.
- 194. Bonacorso, H. G.; Martins, D. B.; Martins, M. A. P.; Zanatta, N.; Flores, A. F. C. Synthesis 2005, 809–813.
- 195. Bonacorso, H. G.; Bittencourt, S. R. T.; Lourega, R. V.; Flores, A. F. C.; Zanatta, N.; Martins, M. A. P. Synthesis 2000, 1431–1434.
- 196. Bozhenkov, G. V.; Frolov, Yu. L.; Toryashinova, D. S.-D.; Levskovskaya, G. G.; Mirskova, A. N. Russ. J. Org. Chem. 2003, 39, 807–813.
- 197. Zanatta, N.; Borchhardt, D. M.; Alves, S. H.; Coelho, H. S.; Squizani, A. M. C.; Marchi, T. M.; Bonacorso, H. G.; Martins, M. A. P. Bioorg. Med. Chem. 2006, 14, 3174–3184.
- 198. Bonacorso, H. G.; Drekener, R. L.; Rodrigues, I. R.; Vezzosi, R. P.; Costa, M. B.; Martins, M. A. P.; Zanatta, N. J. Fluorine Chem. 2005, 126, 1384–1389.
- 199. Bonacorso, H. G.; Duarte, S. H. G.; Zanatta, N.; Martins, M. A. P. Synthesis 2002, 1037–1042.
- 200. Dade, J.; Provot, O.; Moskowitz, H.; Mayrargue, J.; Prina, E. Chem. Pharm. Bull. 2001, 49, 480–483.
- 201. Boltacheva, N. S.; Filyakova, V. I.; Charushin, V. N. Russ. Chem. Bull. 2005, 41, 1452–1457.
- 202. Masquelin, T.; Obrecht, D. Tetrahedron 1997, 53, 641–646.
- 203. Bonacorso, H. G.; Marques, L. M. L.; Zanatta, N.; Martins, M. A. P. Synth. Commun. 2002, 32, 3225–3232.
- 204. Cocco, M. T.; Congiu, C.; Onnis, V.; Bernard, A. M.; Piras, P. P. J. Heterocycl. Chem. 1997, 34, 1347–1350.
- 205. Arun Dutt, S. V.; Chalapathi Rao, C. V. J. Fluorine Chem. 1996, 79, 7–8.
- 206. Desenko, S. M.; Gladkov, E. S.; Nenaidenko, V. G.; Shishkin, O. V.; Shishkinam, S. V. Chem. Heterocycl. Compd. 2004, 40, 65–69.
- 207. Pryadeina, M. V.; Burgart, Ya. V.; Saloutin, V. I.; Kodess, M. I.; Ulomskii, E. N.; Rusinov, V. L. Russ. J. Org. Chem. 2004, 40, 902–907.
- 208. Kuznetsova, O. A.; Filyakova, V. I.; Pashkevich, K. I.; Ulomskii, E. N.; Plekhanov, P. V.; Rusinov, G. L.; Kodess, M. I.; Rusinov, V. L. Russ. Chem. Bull. 2003, 52, 1190–1194.
- 209. Takahashi, M.; Nagaoka, H.; Inoue, K. J. Heterocycl. Chem. 2004, 41, 525–528.
- 210. Zhu, S.-Z.; Qin, C.-Y.; Wang, Y.-L.; Chu, Q.-L. J. Fluorine Chem. 1999, 99, 183–187.
- 211. Kacharova, L. M.; Gerus, I. I.; Kacharov, A. D. J. Fluorine Chem. 2002, 117, 193–197.
- 212. Krasovsky, A. L.; Nenajdenko, V. G.; Balenkova, E. S. Synthesis 2002, 1379–1384.
- 213. Krasovsky, A. L.; Hartulyari, A. S.; Nenajdenko, V. G.; Balenkova, E. S. Synthesis 2002, 133–137.
- 214. Krasovsky, A. L.; Moiseev, A. M.; Nenajdenko, V. G.; Balenkova, E. S. Synthesis 2002, 901–905.
- 215. Krasovsky, A. L.; Moiseev, A. M.; Nenajdenko, V. G.; Balenkova, E. S. Chem. Heterocycl. Compd. 2002, 2, 231– 237.
- 216. Nikishin, K. G.; Nesterov, V. N.; Kislyi, V. P.; Shestopalov, A. M.; Semenov, V. V. Russ. Chem. Bull. 1998, 47, 679–681.
- 217. Krasovsky, A. L.; Moiseev, A. M.; Nenajdenko, V. G.; Balenkova, E. S. Chem. Heterocycl. Compd. 2004, 40, 667– 675.
- 218. Baraznenok, I. L.; Nenajdenko, V. G.; Balenkova, E. S. Chem. Heterocycl. Compd. 2003, 39, 776–779.
- 219. Tsvetkov, N. P.; Koldobskii, A. B.; Godovnikov, I. A.; Kalinin, V. N. Dokl. Akad. Nauk 2005, 404, 785–787.
- 220. Krasovsky, A. L.; Nenajdenko, V. G.; Balenkova, E. S. Tetrahedron 2001, 57, 201–209.
- 221. Krasovsky, A. L.; Pissarev, S. A.; Nenajdenko, V. G.; Balenkova. J. Chem. Soc., Perkin Trans. 1 2002, 2554– 2560.
- 222. Asao, N.; Asano, T.; Yamamoto, Y. Angew. Chem., Int. Ed. 2001, 40.
- 223. Koldobsky, A. B.; Shilova, O. S.; Kalinin, V. N. Mendeleev Commun. 2001, 11, 99–100.
- 224. Obrecht, D.; Zumbrunn, C.; Muller, K. J. Org. Chem. 1999, 64, 6182–6189.
- 225. Volochnyuk, D. M.; Kostyuk, A. N.; Sibgatulin, D. A.; Chernega, A. N.; Pichnuk, A. M.; Tolmachev, A. A. Tetrahedron 2004, 60, 2361–2371.
- 226. Andrew, R. J.; Mellor, J. M.; Reid, G. Tetrahedron 2000, 56, 7255–7260.
- 227. Nenaidenko, V. G.; Druzhinin, S. V.; Balenkova, E. S. Russ. Chem. Bull. 2004, 53, 435–442.
- 228. Reddy, M. V. R.; Rudd, M. T.; Ramachandran, P. V. J. Org. Chem. 2002, 67, 5382–5385.
- 229. Nenajdenko, V. G.; Druzhinin, S. V.; Balenkova, E. S. Mendeleev Commun., in press.
- 230. Nenajdenko, V. G.; Druzhinin, S. V.; Balenkova, E. S. Tetrahedron Lett. 2005, 61, 8853–8885.
- 231. Nenajdenko, V. G.; Druzhinin, S. V.; Balenkova, E. S. Russ. J. Org. Chem., in press.
- 232. Mellor, J. M.; El-Sagheer, A. H.; Salem, E. E.-D. M. Tetrahedron Lett. 2000, 41, 7383–7386.
- 233. Coles, S. J.; Mellor, J. M.; El-Sagheer, A. H.; Salem, E. E.-D. M.; Metwally, R. N. Tetrahedron 2000, 56, 10057–10066.
- 234. Mellor, J. M.; El-Sagheer, A. H.; El-Tamany, E.-S. H.; Metwally, R. N. Tetrahedron 2000, 56, 10067–10074.
- 235. Mellor, J. M.; Reid, G.; El-Sagheer, A. H.; El-Tamany, E.- S. H. Tetrahedron 2000, 56, 10039–10055.
- 236. Zhu, S.; Qin, C.; Xu, G.; Chu, Q.; Huang, Q. J. Fluorine Chem. 1999, 99, 141–144.
- 237. Mellor, J. M.; El-Sagheer, A. H. Tetrahedron Lett. 2000, 41, 7387–7390.
- 238. Singh, R. P.; Kirchmeier, R. L.; Shreeve, J. M. Org. Lett. 1999, 1, 1047–1049.
- 239. Zhu, S.; Jiang, H.; Jin, G. J. Fluorine Chem. 2005, 126, 931– 936.
- 240. Chen, Y.; Huang, L.; Zhang, X. P. J. Org. Chem. 2003, 68, 5925–5929.
- 241. Lee, M.-Y.; Huang, L.; Zhang, X. P. Organometallics 2003, 22, 4905–4909.
- 242. Bryce, M. R.; Chalton, M. A.; Chesney, A. C.; Catterick, D.; Yao, J. W.; Howard, J. A. K. Tetrahedron 1998, 54, 3919– 3928.
- 243. Volle, J.-N.; Schlosser, M. Eur. J. Org. Chem. 2002, 1490– 1492.
- 244. Okano, T.; Matsubara, H.; Kusukawa, T.; Fujita, M. J. Organomet. Chem. 2003, 676, 43–48.
- 245. Nenajdenko, V. G.; Smolko, K. I.; Balenkova, E. S. Tetrahedron: Asymmetry 2001, 12, 1259–1266.
- 246. Yong, K. H.; Chong, J. M. Org. Lett. 2002, 4, 4139–4142.
- 247. Dekeyser, M. A.; Davis, R. A. J. Agric. Food Chem. 1998, 46, 2827–2829.

Biographical sketch

Sergey V. Druzhinin was born in Domodedovo (Moscow region, Russia) in 1980. He graduated from the Department of Chemistry of Moscow State University in 2003. In 2006, he received his Ph.D. degree in organic chemistry from the Department of Chemistry of Moscow State University where he worked under the direction of Professor Valentine G. Nenajdenko. He is currently a post-doctoral fellow in Professor Valentine G. Nenajdenko research group. His research interests include organic synthesis, focusing on heterocyclic and fluorine-containing compounds.

Elizabeth S. Balenkova was born in Moscow in 1926. She graduated from Moscow State University in 1950 and then she was a postgraduate student of the Department of Chemistry of Moscow State University. She received her Ph.D. degree under the supervision of academician B. A. Kazansky in 1953 for the research concerning medium ring hydrocarbons. Since that, she has been working at Moscow State University as a senior researcher (1959) and full professor (1986). She was a supervisor of 27 postgraduate and 63 diploma works. Her research interests are in the area of organic synthesis, electrophilic addition reaction, chemistry of heterocyclic and sulfur compounds.

Valentine G. Nenajdenko was born in 1967 in Ivanovo, Russia. He graduated from Moscow State University (Lomonosov) in 1991. He received his Ph.D. degree under the supervision of Professor E. S. Balenkova in 1994 researching the synthesis and application of unsaturated CF₃-ketones. In 2000 he received Dr. of Chemistry degree concerning chemistry of sulfonium and iminium salts. In 2003, he became full Professor of Organic Chemistry at the Department of Chemistry of Moscow State University. The field of his scientific interest includes organic synthesis, asymmetric catalysis, chemistry of sulfur and fluorine-containing compounds, and chemistry of various heterocycles. He was a supervisor of nine postgraduate works. Valentine G. Nenajdenko is head of Scientific Committee and Jury of International Mendeleev Chemistry Olympiad. He was the winner of the Academiae Europeae Award in 1997, the Russian President Award in 1996, the Prize for the best scientific work at the Department of Chemistry of Moscow State University 2001, Shuvalov Award 2001, the Russian President Award in 2004, Russian Science Support Foundation in 2005.